

Efficiency of Interfacial Electron Transfer from Zn-Porphyrin Dyes into TiO_2 Correlated to the Linker Single Molecule Conductance

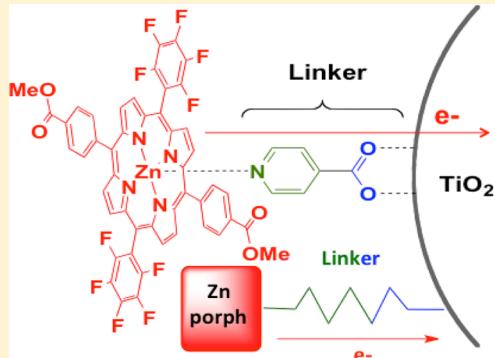
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Supporting Information

ABSTRACT: High performance dye-sensitized solar cells (DSSCs) rely upon molecular linkers that allow efficient electron transport from the photoexcited dye into the conduction band of the semiconductor host substrate. We have studied photoinduced electron injection efficiencies from modular assemblies of a Zn-porphyrin dye and a series of linker molecules which are axially bound to the Zn-porphyrin complex and covalently bound to TiO_2 nanoparticles. Experimental measurements based on terahertz spectroscopy are compared to the calculated molecular conductance of the linker molecules. We find a linear relationship between measured electron injection efficiency and calculated single-molecule conductance of the linker employed. Since the same chromophore is used in all cases, variations in the absorptivities of the adsorbate complexes are quite small and cannot account for the large variations in observed injection efficiencies. These results suggest that the linker single-molecule conductance is a key factor that should be optimized for maximum electron injection efficiencies in DSSCs. In addition, our findings demonstrate for the first time the possibility of inferring values of single molecule conductance for bridging molecules at semiconductor interfaces by using time-resolved THz spectroscopy.



1. INTRODUCTION

Ever since the seminal work of O'Regan and Grätzel,¹ dye-sensitized solar cells (DSSCs) have continued to raise significant interest as promising alternatives for solar-to-electric energy conversion. DSSCs are particularly attractive since they are based on inexpensive semiconductor materials (typically nanoporous TiO_2 , ZnO , or SnO_2)^{2–5} sensitized to visible light absorption with dyes covalently attached to the semiconductor surface by molecular linkers. Photoexcitation of the sensitizer dye leads to electron–hole pair separation, injecting electrons through the molecular linkers into the conduction band (CB) of the semiconductor surface. The photogenerated electron carriers then flow through the external load toward the cathode, performing useful electrical work along the way. At the cathode, electrons reduce a redox mediator (e.g., I_3^-), generating a redox species (e.g., I^-) which refills the holes left behind on the sensitizer molecules.^{2,6} While significant research effort has been devoted to the study of DSSCs, the outstanding challenge is to improve the photoconversion efficiency. Here, we focus on the transport properties of the molecular linkers as directly correlated to the efficiency of electron injection into the semiconductor.

A variety of dyes for DSSCs have been explored beyond the initially utilized Ru–polypyridyl complexes, including inexpensive alternatives such as π -conjugated organic molecules that bypass the need of precious metals.^{7–11} These organic dyes are promising because they exhibit large $\pi-\pi^*$ excitation cross

sections and absorb strongly in the visible region of the spectrum.^{12–18} Organic DSSCs with efficiencies up to 11% have been prepared very recently.¹⁹ Zn-porphyrin dyes, which have been incorporated in some of the most efficient DSSCs to date,^{20,21} are particularly attractive since they have high molar absorptivity. In addition, the spectral and electronic properties can be tuned by varying both the centrally coordinated metal ion and the peripheral substituents.^{22–24} Metal porphyrins can bind directly to the semiconductor surface via built-in anchoring groups at the beta- and meso-positions or alternatively through linkers that coordinate to the metal center in the axial position (Figure 1a).^{25–28} The advantage of the axial binding motif is that it separates the design and synthesis of the linker from the development of the porphyrin dye, thereby simplifying the design of the chromophore/redox active center and assembly of the molecular framework to the semiconductor surface. Nevertheless, axial linkers have been much less investigated, compared to anchoring groups substituting at the beta or meso positions of the porphyrin ring. It is therefore important to analyze the efficiency of photoinduced electron injection as a function of the linker structure²⁵ and explore structure/function relations that might

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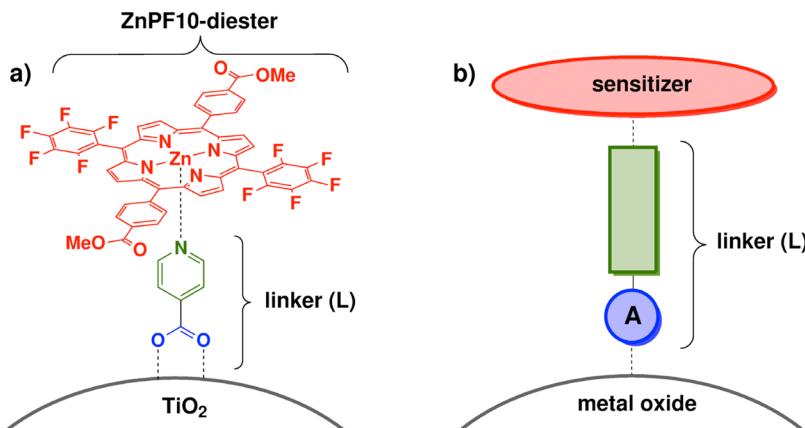


Figure 1. (a) ZnPf_{10} -diester bound to TiO_2 via a pyridyl linker using the modular assembly technique. (b) Cartoon depiction of the modular assembly method for attaching Zn porphyrins to TiO_2 via simple molecular linkers with varying anchor groups. The porphyrin sensitizer, linker (L), and anchors (A) are described in the text.

correlate the resulting interfacial electron transfer efficiency with the electrical properties of the linkers.

High-potential porphyrin dyes have been assembled in photoanodes for DSSCs²⁹ as well as in photocatalytic cells for light-driven water splitting when codeposited with a water oxidation catalyst on TiO_2 thin films.³⁰ They were found to be quite versatile since they could be optimized to give sufficiently positive redox potentials, as necessary to activate catalysts for water oxidation. Furthermore, when covalently bound to semiconductor surfaces, they were stable under oxidative and aqueous conditions, as necessary for photocatalytic water oxidation. However, the reported cells have shown low efficiency calling for fundamental studies of the factors responsible for efficiency loss.

In addition to the optimization of the spectral and electrochemical properties of the dye, optimum performance requires assemblies that maximize electron injection into the semiconductor.^{31–35} For metal-porphyrins bound through axial coordination, electron injection can be optimized through suitable design and refinement of the molecular linker.³⁶ Two photoinjection mechanisms are possible: photoexcitation of the dye, followed by injection through the molecular linker (type-I), or direct injection from the ground state of the dye to the CB of the semiconductor (type-II).^{37–40} While both mechanisms involve ultrafast time scales,^{41,42} the type-I process is usually favored in porphyrin dyes as well as in the molecular assemblies most commonly used in DSSCs.⁴³ Therefore, it is important to focus on optimization of interfacial electron injection by design of molecular linkers that provide proper alignment of electronic states and strong electronic coupling with the semiconductor CB.

Several experimental techniques have been used to study photoinjection mechanisms in DSSCs and to determine the efficiency and time scales of interfacial electron transfer at a variety of semiconductor surfaces, including transient absorption spectroscopy, electron paramagnetic resonance, and time-resolved terahertz (THz) spectroscopy.^{5,44–50} In addition, analytical studies and computational simulations of interfacial electron transfer have advanced our understanding of the complex interrelation between the dye and the semiconductor surface.^{23,40,51–59} These studies have shown that highly conductive linkers, such as molecular frameworks with conjugated double bonds, usually inject carriers much faster than aliphatic linkers. In addition, several studies have focused

on the differences in electron injection observed for different anchoring groups.^{25,60–63} However, a systematic study of the molecular origin of differences in the efficiency of interfacial electron injection has yet to be reported. In this work, we employ a series of modular linker-porphyrin assemblies that vary in their anchoring group for surface attachment. We find that the efficiency of interfacial electron injection as probed by THz spectroscopy correlates with the single molecule conductance of the linkers binding Zn-porphyrin dyes to TiO_2 surfaces. The resulting insight is particularly relevant to the design of linker-porphyrin assemblies when combined with optimization of photoabsorption and redox properties by inverse design techniques.³⁶ These findings also demonstrate for the first time the possibility of inferring values of single molecule conductance by using time-resolved THz spectroscopy for molecules bridging electron donor adsorbates at semiconductor interfaces.

2. EXPERIMENTAL AND COMPUTATIONAL METHODS

2.1. Sample Preparation. All compounds were obtained or synthesized from commercially available starting materials and were used without further purification unless otherwise noted. Isonicotinic acid (L1) and 4-pyridyl-boronic acid (L5) were obtained from Aldrich. The molecular linkers 4-pyridyl-phosphonic acid (L2), 4-pyridyl-acetylacetone (L3), and 4-pyridyl-hydroxamic acid (L4) were synthesized according to a previously reported literature procedure.^{64–68} The high-potential zinc porphyrin dye ZnPf_{10} -diester was also synthesized according to a previously reported literature procedure.^{29,30}

Mesoporous thin films of TiO_2 were prepared using Degussa Aerioxide P25 titanium dioxide nanoparticles by either doctor-blading or spin-coating. Films used for absorbance measurements were prepared on microscope coverslips (25×25 mm for spin-coating, 22×50 mm for doctor-blading, Fisherbrand, USA). Films used for time-resolved THz spectroscopy measurements were prepared on fused quartz microscope slides (1×1 in, 1 mm thickness, GM Associates, Inc., USA). Bare TiO_2 thin films were soaked overnight in the dark in a 0.5 mM solution of the desired linker (L1-L5) in ethanol and thoroughly rinsed with ethanol to remove unbound linker. The films were dried at room temperature for at least 30 min, soaked overnight in the dark in a 0.1 mM dichloromethane solution of the ZnPf_{10} -diester porphyrin, and rinsed well with

dichloromethane to remove any uncoordinated or aggregated porphyrin.

Absorption spectra were recorded using a Cary 3E UV-visible spectrophotometer (Agilent Technologies). Thin films of TiO_2 were highly scattering so absorbance spectra of all thin films were taken in diffuse reflectance geometry using an integrating sphere attachment.

2.2. Time-Resolved THz Spectroscopy (TRTS) Measurements. An amplified Ti:sapphire laser (Tsunami/Spitfire from Spectra Physics) generated 800 mW of pulsed near-IR light at a 1 kHz repetition rate. The pulse width was \sim 130 fs, and the center wavelength was 800 nm. Roughly two-thirds of the power was frequency doubled and then filtered to produce 40 mW of 400 nm (3.1 eV) light for the pump beam. The remainder of the near-IR light was used to generate THz radiation via optical rectification in a ZnTe(110) crystal and detect it using free space electro optic sampling in a second ZnTe(110) crystal. Terahertz data were taken at room temperature, and the average of two samples was taken for each data set. Further information on the spectrometer and technique can be found in the literature.^{45,69–71}

2.3. Conductance and Electronic Structure Calculations. The steady state current flowing through the molecular linker has been computed according to the Landauer–Büttiker formula^{72,73}

$$I = \frac{2e}{h} \int_{\mu_L}^{\mu_R} T(E) dE \quad (1)$$

where e is the electron charge, h is Planck's constant, μ_L and μ_R are the chemical potentials for the electron in the left and right leads, respectively (Figure 2), and E is the energy of the

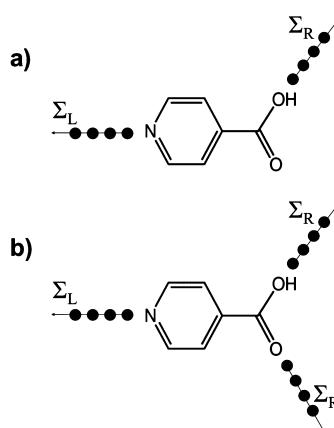


Figure 2. Scheme depicting the coupling of 4-pyridyl-carboxylic acid to the semi-infinite leads in monodentate (a) and bidentate (b) binding motifs.

conducting channel. The transmission function T is calculated by using the Fisher–Lee formula⁷⁴

$$T = \text{Tr}(g_{\text{dev}}^+(E)\Gamma_L(E)g_{\text{dev}}^-(E)\Gamma_R(E)) \quad (2)$$

where $g_{\text{dev}}^+(E)$ is the advanced nonequilibrium Green's function (NEGF) operator for the system consisting of the device attached to one-dimensional leads, as shown in Figure 2.⁷⁵

The NEGF operator can be written, as follows:

$$g_{\text{dev}}(z) = (zS - H_{\text{dev}} - \Sigma_L(z) - \Sigma_R(z))^{-1} \quad (3)$$

where z is the generalized complex energy variable, S is the overlap matrix, H_{dev} is the Hamiltonian of the molecule, and Σ_L and Σ_R are the self-energy operators corresponding to the left and right leads, respectively. Γ is defined as follows:

$$\Gamma_{L/R} = i(\Sigma_{L/R}^+(E) - \Sigma_{L/R}^-(E)) \quad (4)$$

where $\Sigma_{L/R}^+(E)$ and $\Sigma_{L/R}^-(E)$ are the advanced and retarded self-energy operators for the left (L) and right (R) contacts, respectively.

The electronic structure and transport properties of the linkers were determined by using the semiempirical extended-Hückel (EH) Hamiltonian in conjunction with the NEGF methodology.^{76–78} In the basis of localized atomic orbitals, the matrices $\Sigma_{L/R}$ are assumed to be diagonal with nonzero elements $(\Sigma_{L/R})_{11} = \gamma^2 g_0^{+L/R}(E)$ and $(\Sigma_{L/R})_{NN} = \gamma^2 g_0^{+L/R}(E)$ where the indices $1 - N$ label the atomic orbitals of the linker that have significant overlap with the left and right leads. The parameter γ defines the effective electronic coupling with the contacts and is assumed to be the same for s, p, and d orbitals. The Green's function for the isolated left and right contacts, $g_0^{+L/R} = -i/|\beta|$, are defined to mimic semi-infinite one-dimensional nanowires as described by a nearest-neighbor tight-binding Hamiltonian with constant intersite coupling β .⁷⁹ The value of $\gamma^2/|\beta| = 2.0$ eV was chosen to ensure that the conductance of a linear chain of gold atoms at 0 V bias is equal to the unit of quantum conductance (i.e., $G_0 = 2e^2/h$) when connected to the model nanowire leads, as recently reported.⁷⁸ All calculations of conductance were performed at 0 V bias. Hence

$$G = \frac{2e^2}{h} T(E_F) \quad (5)$$

where G is the molecular conductance (in Siemens), and the transmission function T is evaluated at the device Fermi level E_F . The geometries of the linkers (Figure 3) correspond to optimized minimum energy configurations, obtained via density functional theory (DFT) at the B3LYP/LANL2DZ level, as implemented in Gaussian 09.⁸⁰

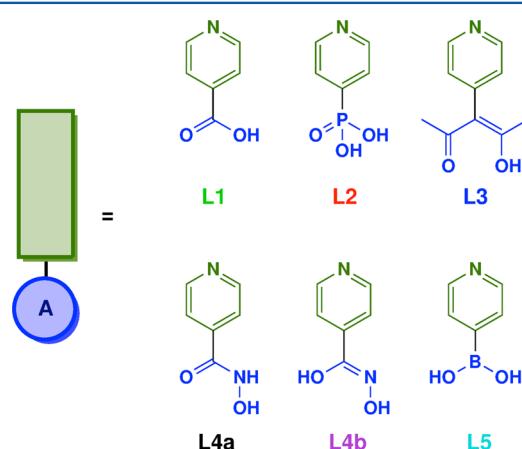


Figure 3. Linker molecules (L) substituted with different anchoring groups (A) that covalently bind Zn-porphyrin sensitizers to TiO_2 surfaces. Linkers have the structure 4-pyridyl-A, where the anchoring group A is either carboxylic acid (L1), phosphonic acid (L2), acetylacetone (L3), hydroxamic acid (L4), or boronic acid (L5).

3. RESULTS AND DISCUSSION

Figure 3 shows the structures of the linkers analyzed in this study. All the linkers have the form 4-pyridyl-A where the anchoring group A is either carboxylic acid (L1), phosphonic acid (L2), acetylacetone (L3), hydroxamic acid (L4), or boronic acid (L5). Hydroxamic acid is treated in both of its two tautomers L4a and L4b.⁸¹ These anchoring groups are known to bind strongly to semiconductor surfaces while the pyridyl group axially coordinates to the metal center of the porphyrin ZnPF₁₀ through the pyridyl nitrogen, sensitizing the TiO₂ surface (Figure 1).⁸² These surface-bound complexes are collectively denoted L-ZnPF₁₀, with L = L1–L5.

Figure 4 shows the UV-vis diffuse reflectance spectra of the L-ZnPF₁₀ assemblies bound to TiO₂ showing only small

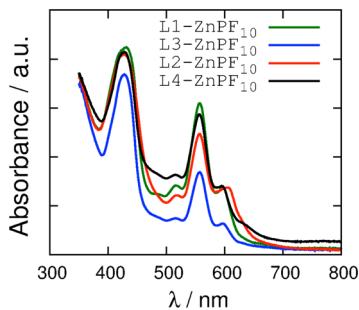


Figure 4. Experimental UV-vis diffuse reflectance spectra for L-ZnPF₁₀ complexes covalently bound to nanoporous TiO₂ thin films.

differences in the molar absorptivity of the chromophore-linker assemblies. In fact, Figure 4 shows that L1-ZnPF₁₀, L2-ZnPF₁₀, and L4-ZnPF₁₀ have nearly the same absorbance at around 400 nm, which is the wavelength of the pump pulse in our THz measurements, while the absorptivities of L3-ZnPF₁₀ (shown) and L5-ZnPF₁₀ (not shown) are only slightly lower. Therefore, differences in the observed photoinduced yields of these assemblies cannot be attributed to differences in light-harvesting efficiency.

Figure 5 shows the change in transmitted THz amplitude as a function of time after the TiO₂ thin-films sensitized with L-ZnPF₁₀ are photoexcited with a pump pulse at 400 nm. The attenuation of THz amplitude is a measure of the efficiency of electron injection, since it is proportional to the concentration

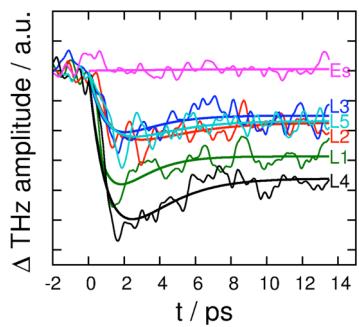


Figure 5. Change in THz amplitude (thin lines) measured as a function of time after photoexcitation at 400 nm for L-ZnPF₁₀ assemblies covalently bound to TiO₂ surfaces as shown in Figure 1. Thick lines are best fitting curves determined by the kinetic model described in the text (eq 6). As a control, the magenta corresponds to ZnPF₁₀-diester (Es) physisorbed to the TiO₂ surface in the absence of a molecular linker.

of free charge carriers injected into the semiconductor conduction band (denoted state B in Figure 6).^{45,71,83,84} As a

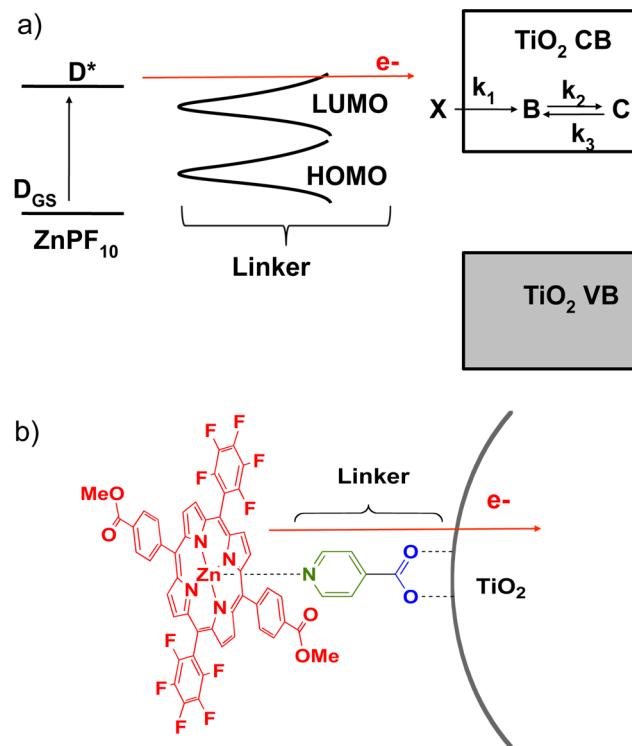


Figure 6. Schematic diagram of photoinduced electron injection in terms of the most relevant energy levels (a) and molecular structures (b). D_{GS} and D* correspond to the porphyrin dye ZnPF₁₀ in the ground (S₀) and second excited state (S₂), respectively. X represents the carrier concentration at the TiO₂ surface prior to injection, B is the concentration of free carriers injected into the TiO₂ conduction band, and C is the concentration of trapped carriers.

control experiment, we note there is no detectable attenuation of THz transmittance when ZnPF₁₀ is merely physisorbed on the TiO₂ surface, whereas all of the L-ZnPF₁₀ complexes covalently bound to the surface show measurable attenuation. We also performed a second control experiment that shows no injection from porphyrin-free L1-sensitized anatase compared to the full system (L1-ZnPF₁₀; see Figure S4). Since the predominant porphyrin excitation near 400 nm (the Soret or B band) arises from the S₂ → S₀ transition, which is π* → π, these results suggest that the photoinduced carriers must travel through the linkers (or through space) in order to be injected into TiO₂. From the THz data alone, however, we see significant differences in the efficiency of electron injection for the different anchoring groups. The efficiencies of electron injection for the various L-ZnPF₁₀ complexes can be ranked according to the order: L4 > L1 > L2 ≈ L3 ≈ L5.

Figure 6 shows a schematic diagram of the photoinduced electron injection process. Upon photoexcitation of the dye, electrons are transferred through the pyridyl ligand and anchoring group to the semiconductor surface (D* → X). This is an ultrafast⁵¹ electron transmission process that takes a few femtoseconds prior to the interfacial injection of free carriers into the conduction band (X → B with an injection rate constant k₁ of tens to hundreds of femtoseconds). The kinetic model also includes trapping (B → C) and detrapping (C → B) dynamics with rate constants k₂ and k₃, respectively. According to this

kinetic model, the concentration of free carriers B , as a function of time t , is given by the following equation:

$$B(t) = X_0(\theta_1 + \theta_2 e^{-k_1 t} + \theta_3 e^{-(k_2+k_3)t}) \quad (6)$$

where

$$\begin{aligned} \theta_1 &= \frac{k_3}{k_2 + k_3} \\ \theta_2 &= \frac{(k_1 - k_3)}{-k_1 + k_2 + k_3} \\ \theta_3 &= \frac{k_1 k_2}{(k_2 + k_3)(k_1 - k_2 - k_3)} \end{aligned} \quad (7)$$

The derivation of eqs 6 and 7 is provided in the Supporting Information (SI). Equation 6 provides a description of the time-dependent concentration of free carriers that should be proportional to the observed attenuation of THz transmitted amplitude for the different anchoring groups. The trapping and detrapping rate constants k_2 and k_3 are assumed to depend only on the nature of the semiconductor and therefore are treated as global fitting parameters common to all anchoring groups. The parameter X_0 , which is reported in Table 1 for each linker, gives

Table 1. Calculated Conductance Values (G) for Linkers in Monodentate and Bidentate Binding Modes and the Maximum Change in THz Amplitude X_0^a

linker	conductance, $G/10^3$ nS		
	bidentate	monodentate	$X_0/\text{arb. units}$
L1	(2.11)	7.26	6.95
L2	(7.90)	3.99	4.56
L3	2.37	(1.16)	3.76
L4-a	13.36	(3.50)	9.31
L4-b	6.27	(2.99)	9.31
L5	(5.77)	2.76	4.29

^aBracketed values do not correlate with X_0 since they do not correspond to the proper binding mode of the ligand (bidentate for L3 and L4 and monodentate for L1, L2 and L5).

a measure of the efficiency of photoinjection as probed by THz spectroscopy since it is proportional to the THz amplitude at the time of maximum attenuation, t_m , and therefore proportional to the concentration of electrons injected into the semiconductor conduction band.

For comparison, Table 1 also reports the values of single molecule conductance computed for each linker by using the EH NEGF methodology outlined in section 2.3. Values for L3 and L4 were calculated assuming a bidentate binding mode, while L1, L2, and L5 were assumed to have monodentate binding. The attachment mode (bidentate for L3 and L4 and monodentate for L1, L2, and L5) is determined for each ligand according to the DFT minimum energy configuration (Figure S5, SI). Our results are consistent with the special arrangement of oxygen atoms, separated by more than two bonds in L3 and L4 anchoring groups, leading to a configuration that is more favorable for bidentate binding with Ti^{4+} octahedral sites.^{49,50} Monodentate binding is found to be most stable for L1, L2, and L5, consistent with previous reports for L1,⁸⁵ L2,⁶¹ and L5.^{86,87} Also, crystallographic data for the dyes absorbed onto novel $(\text{TiO}_2)_{17}$ clusters shows a bidentate chelating mode for L3.⁸⁸

The case of L4 has been previously modeled showing a favorable bidentate oxygen vacancy mode.⁵⁰

The conductance value for L4 was taken as the average between the conductance values for L4a and L4b⁸¹ because the DFT computed energy difference between the two conformers is less than 0.1 eV.

Figure 7 shows the correlation of the maximum change in THz amplitude X_0 as a function of the linker conductance in

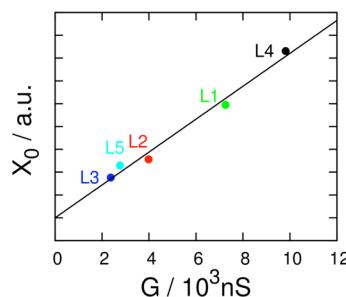


Figure 7. Change in THz amplitude (X_0) at the time of maximum attenuation (t_m) versus conductance G computed for the various linkers. The best linear fit is given by $X_0 = 0.7212G + 1.9943$, with $R^2 = 0.9851$.

the L-ZnPF₁₀ complexes. The linear relationship ($R^2 = 0.9851$), suggests that the transport properties of the different linkers have a direct effect on the electron injection and efficiency. The observed correlation between electron injection efficiency and calculated molecular conductance provides a valuable guideline for designing new sensitizers. To understand why different linkers have different values of conductances, we analyze the electronic structure/function relations for the different linkers as correlated to their transport properties.

Figure 8 displays the frontier molecular orbitals of linkers L1 and L3 that are primarily responsible for electron transport.

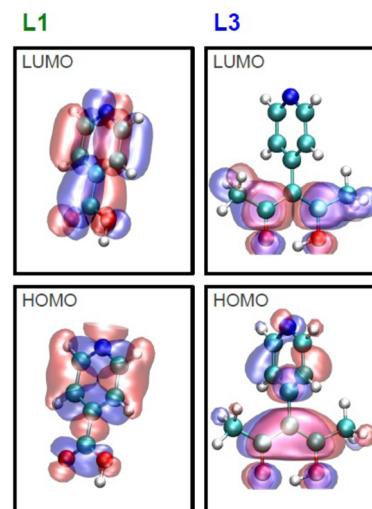


Figure 8. Frontier orbitals calculated within the EH level of theory for L1 and L3.

(Analogous frontier orbitals for L2, L4, and L5 are provided in the SI, Figure S2.) Although L1 and L3 are both known to bind well to the TiO₂ surface,^{2,49} L1 has higher conductance and yields much more efficient electron injection than L3. Figure 8 shows that these differences can be attributed to differences in the electronic structure since the pyridine ring of L3 is rotated

nearly perpendicularly to the anchoring group and therefore is not conjugated with the electronic structure of the anchor as in L1. Furthermore, the LUMO orbital of L3 is localized on the anchor and therefore does not provide good electronic coupling with the pyridine moiety. In contrast, the LUMO of L1 is delocalized along the entire linker, providing an efficient channel for electron transport.

The analysis of L2 and L5 suggests that phosphonic acid (L2) and boronic acid (L5) functional groups decrease the overall conductance since those functional groups inhibit electron tunneling when bound to the semiconductor surface. In contrast, the hydroxamate linker L4 binds as a bidentate ligand, leading to a more efficient injection than through the monodentate L1 linker.⁵⁷ When compared to the bidentate L3 ligand, L4 also leads to much more efficient electron injection due to the extended conjugation and delocalization of the LUMO over the entire linker. EH Frontier orbitals for all the ligands are plotted in Figure S2 in the SI.

The $D^* \rightarrow X$ transport is the primary electron transfer event in porphyrin type-I modular assemblies following photoabsorption at 400 nm due to $\pi^* \leftarrow \pi$ photoexcitation. The photoabsorption and $D^* \rightarrow X$ transport events, however, are dynamically decoupled since the transition dipole moment of the Soret band lies in the plane of the porphyrin ring,⁸⁹ which is parallel to the TiO₂ surface. The resulting decoupling with respect to the adiabatic photoinjection⁹⁰ is consistent with the observation that the photoabsorption features remain almost unchanged when using linkers with different electron transport efficiency. Figure 9 shows the projected DOS (pDOS) for two

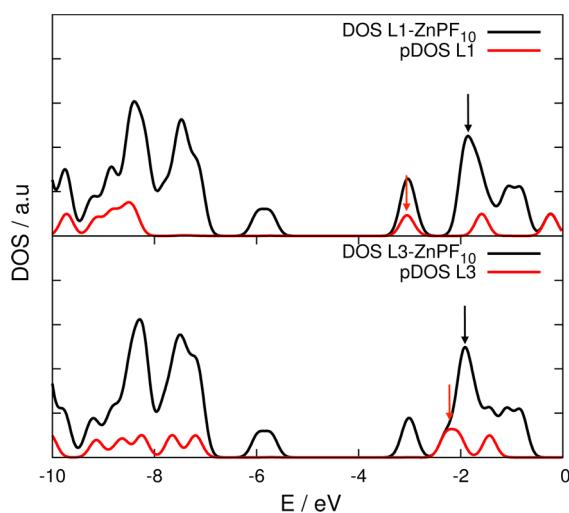


Figure 9. Total density of states (DOS) of the L1-ZnPF₁₀ and L3-ZnPF₁₀ assemblies (black) and projected DOS onto the linkers L1 and L3 (red). The Fermi level is around -4.44 eV. Red arrows indicating the LUMO for the linker projection. Black arrows indicating the final orbitals participating in the $D^* \rightarrow D$ transition.

of the L-ZnPF₁₀ complexes, revealing that the LUMO of the pyridine ligand (red arrows) should be the main orbital responsible for transmission of electrons from the ZnPF₁₀ to the conduction band of TiO₂ as it is located right under the porphyrin projected excited state (black arrows). DOS for all L-ZnPF₁₀ are also shown in Figure S3. These findings suggest that the axial modular assembling technique should be particularly useful for isolating the effect of electron transport properties

during electron photoinjection from the photoabsorption properties of the porphyrin dye.

4. CONCLUSIONS

We have studied the efficiency of interfacial electron injection from Zn-porphyrin dyes bound to TiO₂ surfaces via axially coordinated pyridine ligands with a variety of different anchoring groups. The photoconversion efficiency has been characterized by the attenuation of transmitted time-resolved terahertz (THz) signal after photoexcitation and interfacial electron transfer. We found a linear relationship between the interfacial electron injection efficiency and the single molecule conductance of the linker computed according to the nonequilibrium Green's function formalism. This simple relationship suggests that pyridyl linkers in the surface-bound L-ZnPF₁₀ complexes function as effective molecular wires, transmitting photoexcited electrons from the light-absorbing porphyrin sensitizer into the conduction band of TiO₂. These results are particularly relevant for optimization of DSSCs that rely upon molecular linkers mediating electron transport from the dye into the conduction band of the semiconductor host substrate. In addition, these findings demonstrate for the first time the possibility of inferring values of single molecule conductance by using time-resolved THz spectroscopy to probe bridging molecules that covalently bind dyes to semiconductor surfaces, while keeping constant the other components of the molecular assembly, such as the nature of the dye and the semiconductor material.

■ ASSOCIATED CONTENT

S Supporting Information

Kinetic model and derivation of eq 6; additional figures with the analysis of relaxed configurations, THz analysis; density of states, frontier orbitals, and absorption spectra; coordinates for optimized structures; sample input file for geometry optimizations in SIESTA; and full Gaussian 09 reference. This material is available free of charge via the Internet at <http://pubs.acs.org.org>.

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Notes

The authors declare no competing financial interest.

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Supporting Information for:

**Efficiency of Interfacial Electron Transfer from Zn-Porphyrin Dyes
into TiO₂ Correlated to the Linker Single Molecule Conductance**

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Derivation of B(t) Formula

As shown in Figure 6a of the main text, we consider the injection of an electron from the semiconductor surface (X) to the TiO₂ conduction band (B).



where k_1 is the injection rate constant. In addition, the injected carriers can equilibrate with trap states C ,



where the rate constant for trapping is k_2 and that for detrapping is k_3 . From this, the following kinetic model is obtained:

$$\begin{aligned} \frac{dX}{dt} &= -k_1 X, \\ \frac{dB}{dt} &= +k_1 X - k_2 B + k_3 C, \\ \frac{dC}{dt} &= +k_2 B - k_3 C, \\ X_0 &= X + B + C. \end{aligned} \quad (S3)$$

Solving for X from the first line of Eq. S3 yields $X = X_0 e^{-k_1 t}$, and solving for C , we obtain:

$$\begin{aligned} C &= X_0 - X - B, \\ C &= X_0 - X_0 e^{-k_1 t} - B. \end{aligned} \quad (S4)$$

Inserting the last line of Eq. S4 into the second line of Eq. S3, we obtain:

$$\frac{dB}{dt} = k_1 X_0 e^{-k_1 t} - k_2 B + k_3 X_0 - k_3 X_0 e^{-k_1 t} - k_3 B, \quad (S5)$$

or

$$\frac{dB}{dt} = -(k_2 + k_3)B + k_3 X_0 + X_0 e^{-k_1 t}(k_1 - k_3), \quad (S6)$$

Solving for $B(t)$ we obtain:

$$B(t) = \frac{k_3 X_0}{k_2 + k_3} + \frac{X_0(k_1 - k_3)}{-k_1 + k_2 + k_3} e^{-k_1 t} + \theta_3 e^{-(k_2 + k_3)t} \quad (\text{S7})$$

letting $t = 0$, we find that:

$$\theta_3 = \frac{X_0 k_1 k_2}{(k_2 + k_3)(k_1 - k_2 - k_3)} \quad (\text{S8})$$

We let $\theta_1 = \frac{k_3 X_0}{k_2 + k_3}$ and $\theta_2 = \frac{X_0(k_1 - k_3)}{-k_1 + k_2 + k_3}$ and substitution into Equation S7 yields

$$B(t) = X_0 \left(\theta_1 + \theta_2 e^{-k_1 t} + \theta_3 e^{-(k_2 + k_3)t} \right) \quad (\text{S9})$$

When $k_3 \rightarrow 0$, we recover the familiar solution for consecutive elementary processes $X \xrightarrow{k_1} B \xrightarrow{k_2} C$:

$$B(t) = -\frac{k_1 E_0 (e^{-k_1 t} - e^{-k_2 t})}{k_2 - k_1} \quad (\text{S10})$$

When $k_3 \rightarrow 0$ and $k_2 \rightarrow 0$ we are left with the solution for $E \xrightarrow{k_1} B$:

$$B(t) = E_0 (1 - e^{-k_1 t}) \quad (\text{S11})$$

Values for X_0 , k_1 , k_2 , and k_3 were obtained by performing a nonlinear least squares fit of Equation S9 (Equation 6 in the main text) to the measured transient THz absorption data and are given in Table 2. A global fit was performed wherein k_2 and k_3 were varied simultaneously for all linkers, but each linker had its own independent values of X_0 and k_1 .

Table S1: Fitting parameters for $B(t)$ function. $k_2 = 0.386 \text{ ps}^{-1}$ and $k_3 = 0.253 \text{ ps}^{-1}$.

Linker	X_0 (Arb. Units)	k_1 (ps^{-1})	τ_1 (ps)
L1	6.95	0.70	1.42
L2	4.56	0.58	1.73
L3	3.76	0.83	1.20
L4	9.31	0.70	1.42
L5	4.29	0.65	1.54

Additional figures

The principal electronic transition frequencies and oscillator strengths for L-ZnPF₁₀ dyes were calculated at the TDDFT (B3LYP/LANL2DZ) level of theory. The general spectral features shown in Figure S1 agree with the experimental UV-vis data, as shown in Figure 4 of the main text. From Figure S1, we see that free ZnPF₁₀-diester absorbs at a slightly shorter wavelength than the linker-porphyrin complexes (L-ZnPF₁₀). The red-shift in the porphyrin absorption spectrum upon axial coordination of pyridyl ligands is also observed in the experimental UV-vis and is well-established in the literature. The magnitude of the red-shift was the same for all the linkers (L1-L5) considered in this work. Additionally, we found that L5-ZnPF₁₀ and L3-ZnPF₁₀ have calculated absorptivities at 400 nm that are about 30% larger than for the other complexes. Also, the absorption intensity of L4-ZnPF₁₀ is similar to that of L1-ZnPF₁₀. These observations reinforce that the trends in electron injection efficiency determined by THz spectroscopy (Figure 5 of the main text) cannot be explained by considering experiemntal or calculated absorption intensities alone.

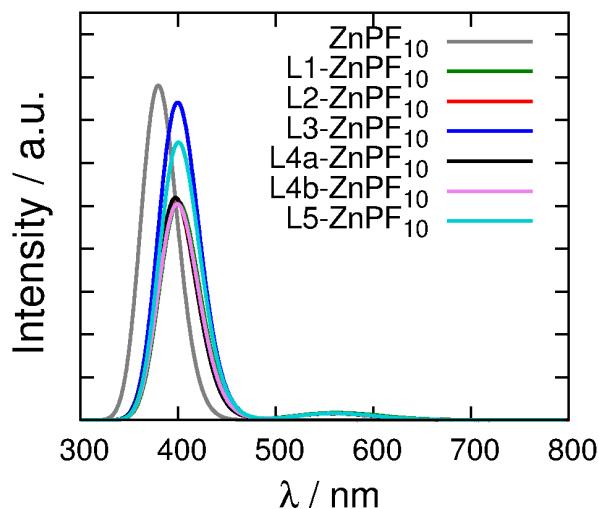


Figure S1: Absorption spectra for L-ZnPF₁₀ complexes. The computed absorption bands have been convoluted with Gaussian functions with a FWHM of 0.4 eV to model the inhomogeneous broadening.

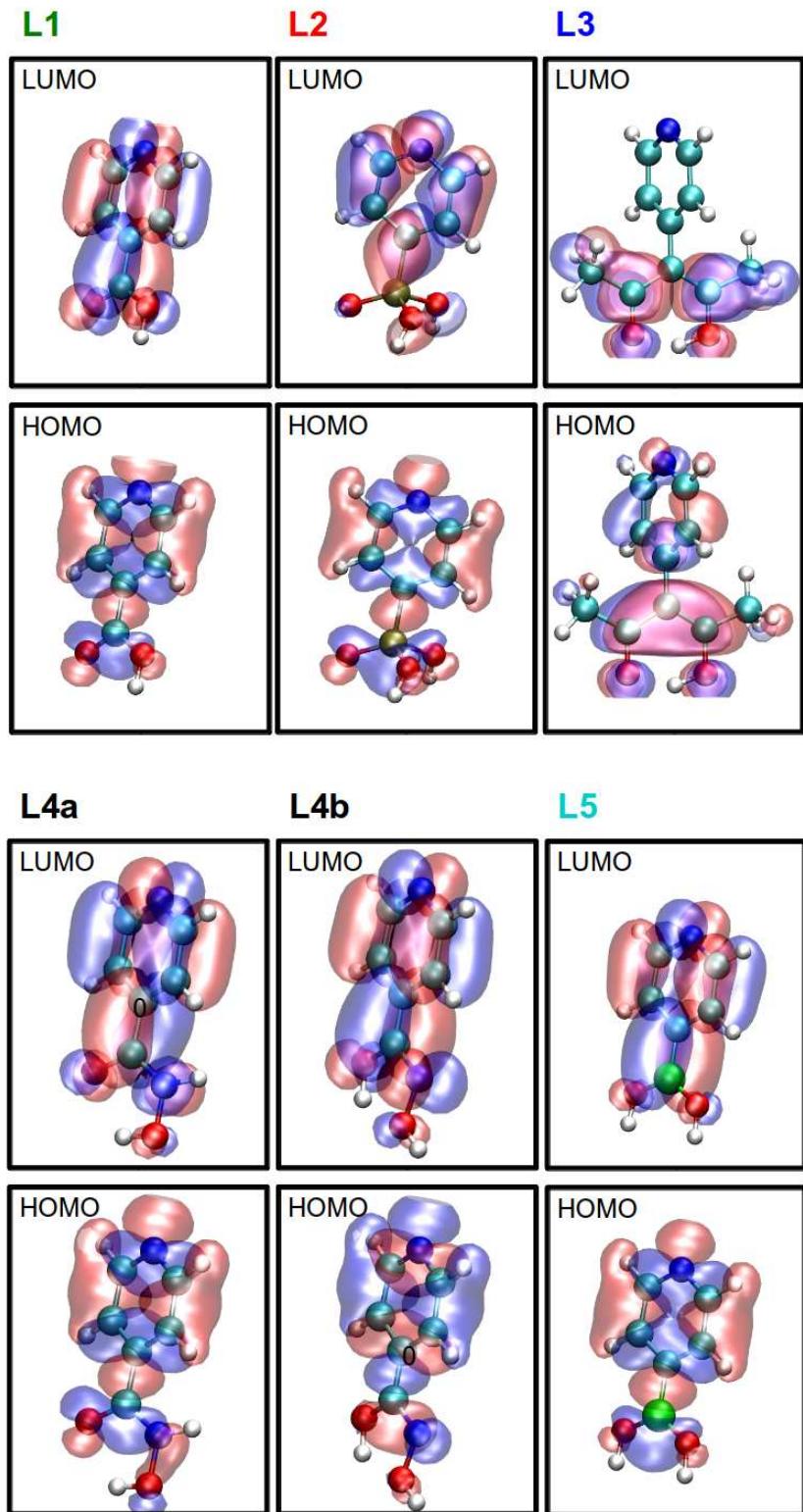


Figure S2: Frontier orbitals calculated within the extended Hückel Hamiltonian for all the linkers considered here.

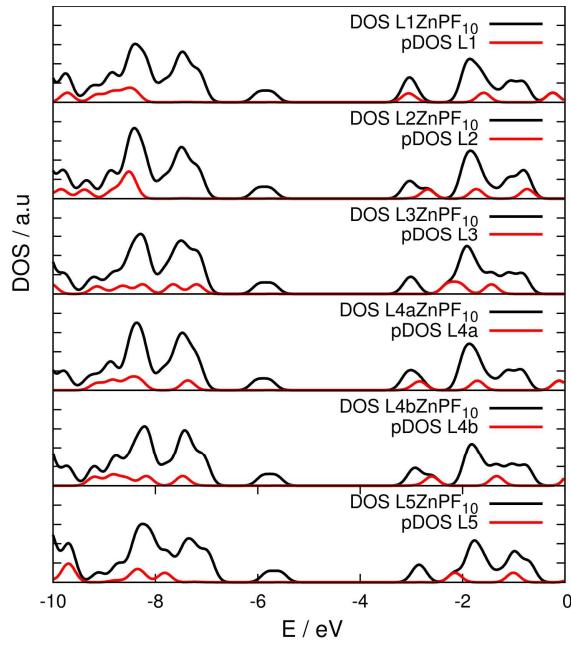


Figure S3: DOS for all the L-ZnPF₁₀ (black line), and DOS projected onto the linkers (pDOS) (red lines). For all the cases the Fermi level of L-ZnPF₁₀ is around -4.44 eV.

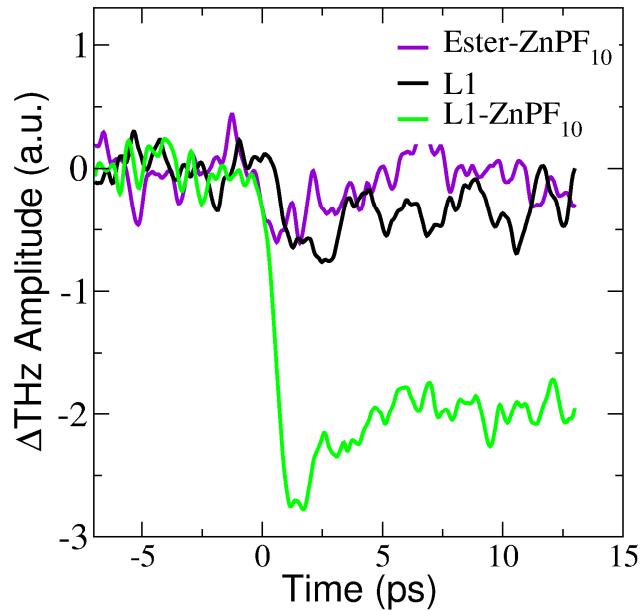


Figure S4: Comparison of sensitization with L1 (only linker) and L1-ZnPF₁₀. The figure shows a big difference in e-injection. The case of just the ligand L1 resembles the Ester-ZnPF₁₀ where hardly any injection is observed.

Geometry optimization for sensitized anatase (101) slab has been performed at the DFT/GGA level of theory by using the SIESTA code with periodic boundary conditions. We used a double zeta (DZ) basis set for all atoms. A confinement radius corresponding to an energy shift of 0.01 eV was employed. A 200 Ry kinetic energy grid cut-off was used. Optimization was done with a force tolerance of 0.04 eV Å⁻¹ fixing the lower layer of Ti atoms and relaxing the rest of the structure. Only the Γ point was used in reciprocal space. Results for L1, L2 and L5 are shown in Figure S4. In all the cases the monodentate mode has a larger binding energy.

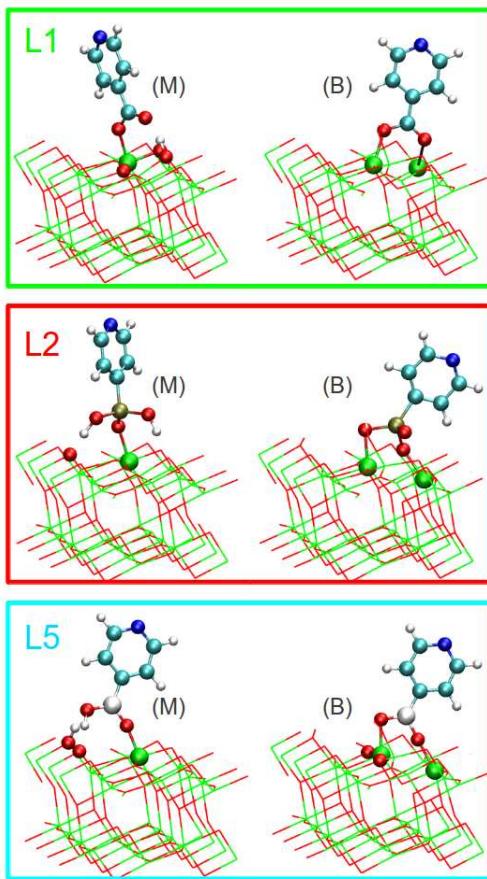


Figure S5: Comparison of relaxed configurations of model nanostructures of TiO₂-anatase sensitized with: L1, L2 and L5 anchored to the (101) surface in a monodentate (M) and a chelate-bitentate (B) mode; The binding energies for L1-M, L1-B, L2-M, L5-M and L5-B are 24, 14, 50, 34 and 23 kcal/mol, respectively. L2-B is predicted to be unstable. Green represents Ti⁴⁺ ions, red represents O²⁻ ions, light blue represents C atoms, deep blue represents N, light brown represents P, large white spheres represent B, and small gray spheres represent H atoms.

Optimized Structures

All optimization were performed using Gaussian09.C01 program, with B3LYP/LANL2DZ level for all atoms.

Table S3-1. Atomic coordinates of L1.

C	1.45847900	-1.71455900	0.00000000
N	0.40740700	-2.57605200	0.00000000
C	-0.84475300	-2.05011400	0.00000000
C	-1.09925100	-0.66626600	0.00000000
C	0.00000000	0.21635400	0.00000000
C	1.30541500	-0.31716900	0.00000000
C	-0.17066000	1.69557900	0.00000000
O	-1.50345600	2.08116300	0.00000000
O	0.75320700	2.52637800	0.00000000
H	-1.58079000	3.06205500	0.00000000
H	2.44556200	-2.16831000	0.00000000
H	-1.66148500	-2.76664500	0.00000000
H	-2.11331800	-0.28349500	0.00000000
H	2.16480400	0.34549400	0.00000000

Table S3-2. Atomic coordinates of L2.

C	-2.50890600	1.15681800	-0.00000400
N	-3.17949900	-0.02159700	-0.00000500
C	-2.45319400	-1.16904600	-0.00000400
C	-1.04552100	-1.19088000	-0.00000200
C	-0.37092400	0.04208300	-0.00000100
C	-1.10213200	1.24191700	-0.00000200
P	1.46353900	0.13719600	0.00000200
O	1.86491700	-0.85227600	-1.33749100
O	2.14706200	1.56952500	-0.00000200
H	-3.11938200	2.05548700	-0.00000500
H	-3.02254500	-2.09419200	-0.00000500
H	-0.50859800	-2.13336800	-0.00000200
H	-0.60266200	2.20626900	-0.00000100
O	1.86491500	-0.85226600	1.33750300
H	2.71274800	-0.67307600	1.79395700
H	2.71275500	-0.67309700	-1.79393900

Table S3-3. Atomic coordinates of L3.

C	0.80015200	2.57584800	-0.30636700
H	0.32408300	2.87714600	0.63584500
H	0.00431800	2.49369700	-1.05540900

H	1.51432300	3.34837700	-0.60166200
C	1.54104700	1.26622200	-0.13586700
C	-2.82546000	0.43873500	1.05800100
N	-3.52712600	-0.04435500	0.00030000
C	-2.81363700	-0.51737100	-1.05364700
C	-1.40673300	-0.52296700	-1.09321800
C	-0.67369600	-0.02146800	0.00684600
C	-1.41937200	0.46813800	1.10367900
C	0.82261400	0.00004000	0.00717300
H	-3.41698600	0.80714400	1.89236400
H	-3.39544900	-0.89842300	-1.88915300
H	-0.89029000	-0.90560100	-1.96937400
H	-0.91555400	0.85242400	1.98647000
C	1.56546400	-1.17943500	0.13023600
O	2.92034300	-1.16149800	0.12372000
C	1.00448200	-2.56467900	0.29274500
H	1.25002400	-3.16871400	-0.59056700
H	-0.07737100	-2.56298300	0.43227900
H	1.48438800	-3.04609900	1.15308000
O	2.83008100	1.29117500	-0.13299000
H	3.23583700	-0.18229800	0.02070100

Table S3-4. Atomic coordinates of L4a.

C	2.46807300	1.00999800	-0.16391800
N	3.04891000	-0.21309800	-0.05102300
C	2.23299900	-1.28481000	0.11523000
C	0.83046200	-1.18575400	0.16546200
C	0.23577000	0.08756300	0.03251500
C	1.07661200	1.20756400	-0.12438500
C	-1.24088300	0.31906300	0.07240100
N	-2.05928900	-0.75157900	-0.15731700
O	-1.76986600	1.44973000	0.30736800
O	-3.47189500	-0.53140100	-0.15127100
H	3.14664100	1.84927800	-0.28782400
H	2.72599200	-2.24775000	0.21793700
H	0.24258500	-2.08348000	0.33673400
H	0.64275300	2.19848300	-0.20644400
H	-3.54350600	0.43747200	0.07181700
H	-1.82592800	-1.66964300	-0.50644200

Table S3-5. Atomic coordinates of L4b.

C	-2.45212900	1.00599000	-0.00015700
N	-3.05135900	-0.21165300	-0.00016400
C	-2.24543000	-1.30759400	-0.00003200
C	-0.84343500	-1.23596800	0.00010800

C	-0.23153300	0.03755700	0.00011000
C	-1.05605800	1.18161200	-0.00002400
C	1.23702800	0.18097000	0.00025400
N	2.01695600	-0.87478000	0.00039800
O	1.71321300	1.47377500	0.00031400
O	3.40566700	-0.39545200	-0.00048500
H	-3.11844700	1.86453000	-0.00026300
H	-2.75196100	-2.26901400	-0.00004100
H	-0.23971900	-2.13704300	0.00021200
H	-0.61794200	2.17310600	-0.00002400
H	3.94301900	-1.21278700	-0.00036700
H	2.69911900	1.46507900	0.00045800

Table S3-6. Atomic coordinates of L5.

C	-1.97256800	1.16120400	-0.00015400
N	-2.67818700	0.00000200	0.000000000
C	-1.97256300	-1.16120700	0.00015400
C	-0.56630000	-1.20690800	0.00015100
C	0.17330300	0.00000300	0.00000100
C	-0.56629900	1.20690600	-0.00015000
B	1.73122800	-0.00000900	0.00000500
O	2.37843800	-1.22569300	-0.00030000
H	-2.56308700	2.07422300	-0.00028000
H	-2.56308700	-2.07422000	0.00028000
H	-0.05096000	-2.16262500	0.00025000
H	-0.05096100	2.16262200	-0.00024900
O	2.37843100	1.22568000	0.00029600
H	3.34542100	1.32642400	0.00027600
H	3.34545400	-1.32628100	-0.00026800

Table S3-7. Atomic coordinates of L1ZnPF₁₀.

N	-0.05977600	0.02820800	1.99838100
C	-0.29865100	-1.11818600	2.69022700
C	-0.35009700	-1.14924200	4.09143100
C	-0.14998400	0.05105200	4.80240900
C	0.09615500	1.23994900	4.08791800
C	0.13449900	1.18455500	2.68591900
C	-0.20700900	0.01921700	6.29346200
O	-0.42586200	-0.99290100	6.97716800
O	0.01369400	1.26368500	6.85843900
H	-0.02837300	1.21947000	7.84053400
H	0.25250700	2.17689900	4.60868300
H	0.32410200	2.07482200	2.09717700
H	-0.45095000	-2.01726900	2.10405000
H	-0.53965700	-2.07587700	4.62236000

C	2.84186700	-1.29963100	-0.52505400
N	1.46261100	-1.48006800	-0.52845300
C	1.22946300	-2.84829400	-0.51795800
C	2.50505000	-3.55805500	-0.48276700
C	3.49589000	-2.60702400	-0.48730200
C	-0.04407300	-3.46201600	-0.54614800
C	-1.29285700	-2.80197400	-0.62084900
C	-2.58371800	-3.46310400	-0.78640300
C	-3.53699800	-2.47586300	-0.85169500
C	-2.84524900	-1.19545400	-0.71368100
N	-1.48040100	-1.42774000	-0.57655800
C	-3.47503000	0.07431600	-0.75209300
C	-2.79927700	1.32022900	-0.71159300
C	-3.45365700	2.62780500	-0.72960100
C	-2.46440500	3.57905400	-0.67588600
C	-1.18946100	2.86913500	-0.62389300
N	-1.42225700	1.50092400	-0.63264600
C	0.08367100	3.48276100	-0.57995500
C	1.33457100	2.82274200	-0.57089500
C	2.63335500	3.48264000	-0.66328400
C	3.58850400	2.49501700	-0.66080200
C	2.88921800	1.21577800	-0.55265800
N	1.51866600	1.44901100	-0.50173900
C	3.51954700	-0.05421500	-0.53970400
H	4.56253900	-2.77207700	-0.45946400
C	-0.07899800	-4.96081100	-0.48825800
H	-2.74248800	-4.52885400	-0.87230800
H	-4.59917600	-2.59927400	-1.00188600
C	-4.97306900	0.09948700	-0.86472500
H	-4.51984600	2.79309200	-0.76992400
H	-2.59280800	4.65224700	-0.67553500
C	0.11497800	4.98211500	-0.53449300
H	2.79760900	4.54755900	-0.74936800
H	4.65794500	2.61750100	-0.74636500
C	5.02152700	-0.08062200	-0.56443700
H	2.63297200	-4.63105000	-0.46047600
Zn	0.00957000	0.01240100	-0.23661800
C	-0.37025000	-5.63207400	0.71179600
C	-0.41463900	-7.02780100	0.80666300
C	-0.15947100	-7.80125400	-0.33217400
C	0.13533400	-7.17051500	-1.54659600
C	0.17151100	-5.77104500	-1.60861900
F	-0.61692500	-4.89315300	1.86129400
F	-0.70333600	-7.64402400	2.00820900
F	-0.19888200	-9.17742600	-0.25719200
F	0.38503900	-7.93004100	-2.67221100
F	0.45751500	-5.18145000	-2.82767400
C	0.32066900	5.66507100	0.67654800

C	0.35782100	7.06161000	0.76103200
C	0.18391400	7.82408400	-0.40027700
C	-0.02397100	7.18154800	-1.62645100
C	-0.05576900	5.78143700	-1.67741900
F	0.56016300	7.68933900	1.97444400
F	0.48624600	4.93751300	1.84802000
F	0.21746900	9.20091400	-0.33550100
F	-0.19349200	7.93000000	-2.77407100
F	-0.25521200	5.18005000	-2.90768700
C	5.71629700	-0.60337400	-1.68060600
C	7.11904300	-0.62467600	-1.71070000
C	7.85624500	-0.12794300	-0.61394900
C	7.17279700	0.39344400	0.50611000
C	5.77103000	0.42024900	0.52622400
C	-5.78350400	-0.38720800	0.18770800
C	-7.18319300	-0.36244000	0.09061600
C	-7.80062200	0.14396800	-1.07344500
C	-7.00106800	0.62609500	-2.13271500
C	-5.60308000	0.60738600	-2.02569600
H	5.24985100	0.81505300	1.39417000
H	7.75202900	0.76851200	1.34492700
C	9.34442300	-0.13288000	-0.59454200
H	7.64782700	-1.01801300	-2.57216900
H	5.15226100	-0.97651200	-2.53112300
H	-4.99253100	0.97006400	-2.84811700
H	-7.48844800	1.00660800	-3.02570200
C	-9.28020600	0.18826000	-1.22768100
H	-7.80122800	-0.72786900	0.90362400
H	-5.31184500	-0.77054400	1.08858500
O	10.04916000	0.29392700	0.34062400
O	9.88937300	-0.67216000	-1.74605300
C	11.35986100	-0.72026100	-1.82345700
H	11.57567900	-1.17647600	-2.79054500
H	11.77406800	0.29108100	-1.76318200
O	-9.88253000	0.62345700	-2.22802900
O	-9.94575400	-0.31382300	-0.12308000
C	-11.41795700	-0.31191600	-0.18101200
H	-11.73749700	-0.73542700	0.77216300
H	-11.78947200	0.71032300	-0.30333000
H	11.76120900	-1.32417600	-1.00360400
H	-11.76303100	-0.92389000	-1.02030100

Table S3-8. Atomic coordinates of L2ZnPF₁₀.

N	-0.05946600	0.01755600	1.77307100
C	-0.10838400	-1.16355900	2.44459000
C	-0.16260800	-1.22861800	3.84594900
C	-0.16392800	-0.02135300	4.56488200

C	-0.11857100	1.20461800	3.88062500
C	-0.06531500	1.17891200	2.47610400
P	-0.22707600	-0.00331500	6.40795800
O	-1.42416600	-1.17908200	6.66769900
O	1.14332900	-0.88990800	6.95053100
H	-0.13094200	2.14982800	4.41452300
H	-0.02912600	2.09875000	1.90342200
H	-0.10528400	-2.06671600	1.84487700
H	-0.21124100	-2.18914200	4.34636100
C	2.81112700	-1.37160100	-0.76229400
N	1.42765600	-1.51454800	-0.72546800
C	1.15513400	-2.87349600	-0.79531000
C	2.40905800	-3.61751600	-0.86866000
C	3.42652300	-2.69450400	-0.85688300
C	-0.13538800	-3.44991400	-0.82771700
C	-1.36569200	-2.75557100	-0.87761700
C	-2.67735300	-3.37917300	-1.02734000
C	-3.60238000	-2.36482100	-1.07679300
C	-2.87214100	-1.10501600	-0.94483100
N	-1.51320900	-1.37669500	-0.82357600
C	-3.46568200	0.18196400	-0.97826800
C	-2.75489400	1.40791400	-0.93365800
C	-3.36982100	2.73417500	-0.96306900
C	-2.35353900	3.65609300	-0.89625800
C	-1.10142400	2.90883000	-0.82317900
N	-1.37421400	1.54801500	-0.83735200
C	0.18781900	3.48433500	-0.75395500
C	1.41917000	2.79011300	-0.73903000
C	2.73742600	3.41717900	-0.71985700
C	3.66567000	2.40437700	-0.72303000
C	2.92922100	1.14131600	-0.74584600
N	1.56446800	1.40978200	-0.74383200
C	3.52372300	-0.14593700	-0.75166800
H	4.48671600	-2.88660800	-0.92808300
C	-0.20484100	-4.94912800	-0.81253800
H	-2.86799200	-4.43986100	-1.11071600
H	-4.66982300	-2.45673200	-1.21144400
C	-4.96259800	0.25043600	-1.09113000
H	-4.42999600	2.93038600	-1.02027500
H	-2.44885300	4.73280100	-0.90256500
C	0.25616200	4.98208900	-0.68221900
H	2.93214300	4.48035800	-0.71055800
H	4.74087100	2.50109500	-0.70455000
C	5.02439000	-0.21871200	-0.76970700
H	2.50404400	-4.69139300	-0.94714300
Zn	0.01553900	0.01764500	-0.46641700
C	-0.22501400	-5.65822000	0.40019500
C	-0.29034100	-7.05518400	0.45812800

C	-0.33797800	-7.78941100	-0.73313100
C	-0.31955300	-7.11945300	-1.96219400
C	-0.25376600	-5.71972000	-1.98583800
F	-0.18082300	-4.95677700	1.59806300
F	-0.30880100	-7.71052900	1.67355400
F	-0.40309200	-9.16608500	-0.69487800
F	-0.36593000	-7.84042500	-3.13854100
F	-0.23538500	-5.08898800	-3.21729500
C	0.21116400	5.65086400	0.55241500
C	0.27458500	7.04496500	0.66019900
C	0.38752100	7.81755800	-0.50215300
C	0.43532000	7.18843000	-1.75185800
C	0.36958700	5.79049400	-1.82519400
F	0.22810700	7.66024200	1.89526900
F	0.10173300	4.90943000	1.72161000
F	0.45210000	9.19196800	-0.41520200
F	0.54712600	7.94764000	-2.89947100
F	0.41925500	5.20052700	-3.07589600
C	5.75711600	0.24158500	-1.88895900
C	7.15833300	0.16863300	-1.91212500
C	7.85599800	-0.36146900	-0.80522700
C	7.13454500	-0.82094300	0.31809700
C	5.73405700	-0.75361600	0.33137600
C	-5.78622300	-0.19512400	-0.03099100
C	-7.18464900	-0.12882700	-0.12815800
C	-7.78702400	0.37868000	-1.29957600
C	-6.97403200	0.81988100	-2.36639900
C	-5.57715600	0.75957300	-2.25959000
H	5.18381700	-1.10176300	1.20139300
H	7.68396800	-1.22292600	1.16446600
C	9.34110100	-0.45612600	-0.77827200
H	7.71599100	0.51389800	-2.77601500
H	5.22285700	0.63992000	-2.74726200
H	-4.95617300	1.09091900	-3.08744800
H	-7.45027800	1.20192500	-3.26472900
C	-9.26474200	0.46663500	-1.45382900
H	-7.81305900	-0.46234800	0.69058100
H	-5.32577600	-0.57859200	0.87564000
O	10.01141600	-0.91522900	0.16686000
O	9.92588700	0.02793800	-1.93466700
C	11.39667500	-0.02428300	-2.00502200
H	11.64642400	0.39416700	-2.98096200
H	11.83408500	0.57035700	-1.19689200
O	-9.85398200	0.90567500	-2.46026500
O	-9.94453600	0.00168700	-0.34179900
C	-11.41600500	0.04954800	-0.39853600
H	-11.74758900	-0.35002900	0.56083200
H	-11.75494200	1.08128400	-0.53491300

H	11.74194000	-1.05935800	-1.91927800
H	-11.78138900	-0.56293400	-1.22881400
H	1.84040300	-0.37640600	7.40949200
H	-1.58229100	-1.49063600	7.58199500
O	-0.36996100	1.42028300	7.08645300

Table S3-9. Atomic coordinates of L3ZnPF₁₀.

N	0.05075800	0.01735900	1.61505700
C	0.13565400	-1.14997200	2.30670500
C	0.17474100	-1.19340500	3.70777200
C	0.12130400	0.00689300	4.45518500
C	0.03321200	1.21309200	3.72073400
C	0.00132500	1.17959700	2.31937700
C	0.16137000	0.00078500	5.94699600
C	-0.84849900	-0.62283300	6.69593300
C	1.26821800	0.63435600	6.66609000
H	-0.02974400	2.16974000	4.22955900
H	-0.07441900	2.09513300	1.74279900
H	0.18053100	-2.06117800	1.72019000
H	0.25925600	-2.15272000	4.20873600
C	2.78431800	-1.30004300	-1.11083600
N	1.41183400	-1.47731300	-0.97168300
C	1.16887400	-2.84223700	-1.01904500
C	2.43171300	-3.55542500	-1.18796100
C	3.42394100	-2.60739100	-1.25300400
C	-0.10558600	-3.45043500	-0.94488100
C	-1.35325300	-2.78630100	-0.91187000
C	-2.65606000	-3.44531300	-0.88453700
C	-3.60831000	-2.45531200	-0.88121400
C	-2.90234200	-1.17465300	-0.90695600
N	-1.53165600	-1.41009200	-0.91093400
C	-3.52907900	0.09741700	-0.91169600
C	-2.84766500	1.34086900	-0.91901800
C	-3.49545700	2.64858700	-1.01353300
C	-2.50108200	3.59653400	-1.01389500
C	-1.22966800	2.88297300	-0.93423600
N	-1.46896900	1.51780300	-0.87413600
C	0.04689200	3.49086700	-0.94534400
C	1.29390600	2.82686600	-0.99675000
C	2.59538400	3.48613400	-1.05899200
C	3.54582500	2.49633300	-1.12097200
C	2.83994200	1.21559300	-1.09867100
N	1.47217100	1.45072600	-1.00824700
C	3.46468000	-0.05655500	-1.14726700
H	4.47999300	-2.77320600	-1.40570600
C	-0.13349800	-4.94997100	-0.89422500
H	-2.82585800	-4.51275900	-0.87464500

H	-4.68075200	-2.57815300	-0.85727200
C	-5.03102300	0.13196900	-0.93131700
H	-4.55951800	2.81491700	-1.09204500
H	-2.62212500	4.66802600	-1.08999300
C	0.07886100	4.99049500	-0.89691100
H	2.76523200	4.55363700	-1.06123600
H	4.61731600	2.61949200	-1.17084000
C	4.96191700	-0.09133100	-1.26871300
H	2.54669900	-4.62662700	-1.27653000
Zn	-0.01743400	0.02060300	-0.60301100
C	0.06298300	-5.63674400	0.31585400
C	0.04507800	-7.03339900	0.40498700
C	-0.17718400	-7.79167500	-0.75080000
C	-0.37798000	-7.14502400	-1.97603000
C	-0.35409200	-5.74485400	-2.03152500
F	0.27670700	-4.91267900	1.48190300
F	0.24060300	-7.66507300	1.61763200
F	-0.19849900	-9.16854800	-0.68148500
F	-0.59600000	-7.88963800	-3.11794300
F	-0.55080700	-5.13910600	-3.25994600
C	-0.00690500	5.67681400	0.32616300
C	0.01927700	7.07347000	0.41344400
C	0.13532700	7.83183500	-0.75776600
C	0.22356800	7.18543300	-1.99636700
C	0.19488700	5.78526200	-2.04948500
F	-0.06543300	7.70504000	1.63881100
F	-0.11564000	4.95220500	1.50632100
F	0.16300300	9.20862700	-0.69035200
F	0.33706200	7.93003200	-3.15328200
F	0.27989800	5.17922100	-3.29044500
C	5.75848900	-0.61268100	-0.22230700
C	7.15743400	-0.64470900	-0.32788700
C	7.78786600	-0.16127100	-1.49474900
C	7.00203700	0.35505000	-2.54815200
C	5.60510200	0.39291700	-2.43258900
C	-5.75085400	-0.35388900	-2.04826000
C	-7.15346200	-0.31833000	-2.07251700
C	-7.86574700	0.19931100	-0.96901400
C	-7.15736200	0.68484400	0.15168300
C	-5.75561100	0.65490500	0.16590100
H	5.00481800	0.78156000	-3.25068100
H	7.49904000	0.71721600	-3.44347100
C	9.26706600	-0.17664700	-1.65795700
H	7.76514700	-1.03696300	0.48041100
H	5.27680600	-0.97916300	0.68030300
H	-5.21529300	1.02317300	1.03383700
H	-7.71781200	1.07734000	0.99528800
C	-9.35282700	0.25366600	-0.94288200

H	-7.70114000	-0.68334500	-2.93466500
H	-5.20571100	-0.74284400	-2.90398500
O	9.88045500	0.23553500	-2.66127700
O	9.91843400	-0.70695000	-0.55803000
C	11.38923800	-0.76385500	-0.62476900
H	11.69720000	-1.20198600	0.32560700
H	11.70465700	-1.38718300	-1.46735100
O	-10.03581000	0.69967800	-0.00052400
O	-9.92383700	-0.25305200	-2.09647200
C	-11.39546700	-0.24117200	-2.16752300
H	-11.63344700	-0.67496400	-3.13970400
H	-11.81709100	-0.83998000	-1.35410400
H	11.80065800	0.24297800	-0.74715400
H	-11.76853800	0.78496000	-2.09128500
C	2.42373400	1.29403100	5.94244100
O	1.29365800	0.62710800	7.95495100
C	-2.06814800	-1.31084100	6.14675600
O	-0.80962700	-0.63601400	8.04646700
H	0.04290600	-0.13465700	8.35934500
H	3.27517700	1.34916200	6.62572800
H	2.15647100	2.31863700	5.65076000
H	2.70582400	0.75593700	5.03199900
H	-2.94224200	-0.99395500	6.72639800
H	-2.23266300	-1.09744900	5.08960200
H	-1.97355900	-2.39721300	6.27660200

Table S3-10. Atomic coordinates of L4aZnPF₁₀.

N	-0.05445000	-0.00551400	1.87819000
C	-0.08220500	-1.19891600	2.52717400
C	-0.13327100	-1.29284000	3.92578100
C	-0.16268400	-0.10661600	4.68936800
C	-0.11865400	1.12942000	4.01451000
C	-0.06859600	1.13879400	2.61286200
H	-0.11803900	2.05400900	4.58102300
H	-0.03863200	2.07209000	2.06182100
H	-0.05606600	-2.09045800	1.91082200
H	-0.11895800	-2.27847500	4.38090900
C	2.81341500	-1.36471600	-0.66445800
N	1.42953900	-1.50360400	-0.63133100
C	1.15288700	-2.86110400	-0.71061200
C	2.40467000	-3.60875400	-0.78623100
C	3.42495100	-2.68885700	-0.76708400
C	-0.13970100	-3.43254600	-0.74874900
C	-1.36754000	-2.73360100	-0.80060800
C	-2.68009800	-3.35156200	-0.96536500

C	-3.60122000	-2.33338300	-1.01498200
C	-2.86772000	-1.07728500	-0.86670500
N	-1.51077600	-1.35482500	-0.73696900
C	-3.45634600	0.21222500	-0.89342600
C	-2.74185000	1.43518300	-0.82912300
C	-3.35188700	2.76387800	-0.85051400
C	-2.33305300	3.68134600	-0.76437500
C	-1.08416800	2.92895000	-0.68789100
N	-1.36169300	1.56948700	-0.71851700
C	0.20677100	3.49981200	-0.60659400
C	1.43584500	2.80147100	-0.59851200
C	2.75659600	3.42338900	-0.56638200
C	3.68090700	2.40743600	-0.58115400
C	2.93961900	1.14749000	-0.62261800
N	1.57601500	1.42087100	-0.62123600
C	3.52987600	-0.14146000	-0.64145700
H	4.48470600	-2.88351400	-0.83798600
C	-0.21520500	-4.93144100	-0.73677700
H	-2.87366100	-4.41077900	-1.06067800
H	-4.66759900	-2.42042200	-1.16103400
C	-4.95165600	0.28754100	-1.02247600
H	-4.41068000	2.96472300	-0.91664500
H	-2.42450800	4.75839400	-0.76185000
C	0.27765800	4.99618500	-0.51379600
H	2.95550900	4.48546100	-0.54023300
H	4.75641000	2.49961200	-0.55767300
C	5.03037600	-0.21859700	-0.65962800
H	2.49664900	-4.68234700	-0.87226500
Zn	0.02188200	0.03157500	-0.35494700
C	-0.26183300	-5.64110300	0.47474400
C	-0.33413900	-7.03758100	0.53172600
C	-0.36156200	-7.77175800	-0.66017400
C	-0.31652500	-7.10158600	-1.88848300
C	-0.24454900	-5.70200600	-1.91081700
F	-0.23762600	-4.94034000	1.67485000
F	-0.37926000	-7.69209100	1.74723900
F	-0.43329900	-9.14801200	-0.62317600
F	-0.34331300	-7.82219400	-3.06543900
F	-0.20077200	-5.07150400	-3.14146200
C	0.15066600	5.65225400	0.72212200
C	0.21196000	7.04486000	0.84894500
C	0.40803200	7.82922100	-0.29427000
C	0.53937900	7.21303900	-1.54440300
C	0.47308300	5.81635400	-1.63732400
F	0.08318800	7.64726400	2.08453300
F	-0.03858900	4.89950700	1.87348100
F	0.47160100	9.20234200	-0.18832200
F	0.73186800	7.98385000	-2.67347200

F	0.60306000	5.23989500	-2.88868500
C	5.76377600	0.24658800	-1.77635900
C	7.16475600	0.16948700	-1.80067600
C	7.86123800	-0.36984800	-0.69752300
C	7.13904700	-0.83353400	0.42357900
C	5.73874100	-0.76184000	0.43810300
C	-5.78980600	-0.16616600	0.02273300
C	-7.18671400	-0.09323800	-0.08999200
C	-7.77324200	0.42951600	-1.26271900
C	-6.94588400	0.87891400	-2.31495700
C	-5.55060300	0.81172100	-2.19241400
H	5.18808900	-1.11214200	1.30698100
H	7.68763600	-1.24164600	1.26754200
C	9.34609400	-0.46985200	-0.67235300
H	7.72310100	0.51868500	-2.66253200
H	5.23027600	0.65232300	-2.63168200
H	-4.91860100	1.14955200	-3.00920900
H	-7.40992000	1.27265500	-3.21462700
C	-9.24869000	0.52535200	-1.43324400
H	-7.82631700	-0.43312000	0.71740900
H	-5.34140300	-0.56148000	0.93033300
O	10.01530300	-0.94071400	0.26766100
O	9.93183400	0.02375800	-1.82425500
C	11.40246900	-0.03192400	-1.89551300
H	11.65325600	0.39669600	-2.86676500
H	11.84169900	0.55236800	-1.08086000
O	-9.82420200	0.97774200	-2.44161700
O	-9.94339800	0.05132800	-0.33423900
C	-11.41397500	0.10604800	-0.40791800
H	-11.75870400	-0.30418900	0.54227500
H	-11.74703600	1.14084300	-0.53524000
H	11.74477300	-1.06888600	-1.82140500
H	-11.77184400	-0.49431300	-1.25021200
C	-0.21269800	-0.08332200	6.18545200
O	0.09531200	0.93579000	6.87647800
N	-0.61181600	-1.21989200	6.82816700
H	-1.01061000	-2.06645600	6.44897500
O	-0.68453700	-1.19748300	8.25411600
H	-0.36837000	-0.28014900	8.48133300

Table S3-11. Atomic coordinates of L4bZnPF₁₀.

N	-0.05271800	0.01937700	1.88457400
C	-0.08986900	-1.15785200	2.56893100
C	-0.12721100	-1.21529100	3.96675600
C	-0.12604800	-0.00683800	4.69869100
C	-0.08781000	1.21238500	3.99152900
C	-0.05207700	1.18098800	2.58923300

H	-0.08575100	2.15688700	4.52218300
H	-0.02205200	2.10023200	2.01497000
H	-0.08880400	-2.06517900	1.97522600
H	-0.15621900	-2.17095200	4.47743200
C	2.80485500	-1.37345400	-0.65671400
N	1.42201100	-1.51761800	-0.61857100
C	1.15061000	-2.87629600	-0.69025800
C	2.40518100	-3.61961100	-0.76580200
C	3.42182300	-2.69582700	-0.75310200
C	-0.13947000	-3.45353300	-0.72269400
C	-1.37042500	-2.76002900	-0.76834400
C	-2.68166700	-3.38487100	-0.91853000
C	-3.60776500	-2.37134600	-0.96410200
C	-2.87815200	-1.11108900	-0.82981900
N	-1.51927500	-1.38176600	-0.71027500
C	-3.47261300	0.17564100	-0.86245600
C	-2.76270500	1.40228200	-0.82098300
C	-3.37876000	2.72824200	-0.85360700
C	-2.36279400	3.65085300	-0.79224100
C	-1.11015700	2.90411000	-0.71860800
N	-1.38228300	1.54349100	-0.72717800
C	0.17900700	3.48072700	-0.65562400
C	1.41082300	2.78736500	-0.64156900
C	2.72895700	3.41556300	-0.62579100
C	3.65758200	2.40329900	-0.62657600
C	2.92122500	1.13974800	-0.64471100
N	1.55667100	1.40746600	-0.64212600
C	3.51640500	-0.14712000	-0.64772500
H	4.48221200	-2.88677600	-0.82488700
C	-0.20775300	-4.95295600	-0.71578900
H	-2.87125900	-4.44555400	-1.00486400
H	-4.67520000	-2.46405500	-1.09848700
C	-4.96972200	0.24256800	-0.97263700
H	-4.43911300	2.92389400	-0.90972000
H	-2.45861500	4.72753400	-0.80295900
C	0.24610900	4.97886500	-0.59100000
H	2.92322100	4.47887900	-0.61930600
H	4.73277100	2.50043800	-0.60882000
C	5.01711500	-0.21925200	-0.66507800
H	2.50095000	-4.69335600	-0.84563500
Zn	0.00881800	0.01405800	-0.33600800
C	-0.22883200	-5.67040100	0.49209800
C	-0.29304900	-7.06792100	0.54002800
C	-0.33854200	-7.79418900	-0.65604000
C	-0.31914600	-7.11584000	-1.88038200
C	-0.25454100	-5.71593900	-1.89429300
F	-0.18676500	-4.97834200	1.69498600
F	-0.31256800	-7.73181400	1.75113400

F	-0.40259600	-9.17141500	-0.62724200
F	-0.36346500	-7.82900400	-3.06182300
F	-0.23528000	-5.07734300	-3.12171100
C	0.17473800	5.65607200	0.63793100
C	0.23571300	7.05096900	0.73733700
C	0.37358100	7.81611800	-0.42704700
C	0.44822400	7.17865700	-1.67109900
C	0.38408400	5.78023600	-1.73645000
F	0.16267600	7.67458300	1.96749300
F	0.04159600	4.92400300	1.81018200
F	0.43601200	9.19138900	-0.34763700
F	0.58414700	7.93047400	-2.82122700
F	0.45936800	5.18286700	-2.98233500
C	5.75043900	0.23784100	-1.78528500
C	7.15163100	0.16488100	-1.80764300
C	7.84886900	-0.36211400	-0.69890000
C	7.12680800	-0.81812300	0.42553100
C	5.72632000	-0.75072900	0.43798800
C	-5.79109700	-0.20838100	0.08710800
C	-7.18975100	-0.14423400	-0.00722300
C	-7.79529300	0.36682100	-1.17555100
C	-6.98477700	0.81333400	-2.24213400
C	-5.58765700	0.75484300	-2.13806100
H	5.17548800	-1.09570000	1.30886200
H	7.67569100	-1.21736200	1.27354800
C	9.33368900	-0.45694300	-0.67130800
H	7.70963700	0.50767800	-2.67230000
H	5.21641800	0.63370800	-2.64487200
H	-4.96868700	1.08982400	-2.96592700
H	-7.46316400	1.19773800	-3.13832600
C	-9.27312200	0.45266200	-1.32684900
H	-7.81612600	-0.48234000	0.81122300
H	-5.32796600	-0.59475900	0.99111200
O	10.00435200	-0.91389800	0.27471100
O	9.91920900	0.02437300	-1.82892500
C	11.38981300	-0.02854800	-1.89820600
H	11.64037700	0.38708900	-2.87517300
H	11.82709900	0.56816200	-1.09150100
O	-9.86558300	0.89461700	-2.33019500
O	-9.95062600	-0.01812700	-0.21534000
C	-11.42208600	0.02652900	-0.27072000
H	-11.75207900	-0.37794000	0.68718500
H	-11.76375200	1.05798500	-0.40252500
H	11.73486800	-1.06345700	-1.80932500
H	-11.78686000	-0.58322400	-1.10329100
C	-0.16334800	-0.00742400	6.17362500
O	-0.15911600	1.23149900	6.77206200
N	-0.19731600	-1.13768500	6.83900700

O	-0.22786100	-0.80988600	8.26495700
H	-0.24894100	-1.67802100	8.71602700
H	-0.18361700	1.13415000	7.75277900

Table S3-12. Atomic coordinates of L5ZnPF₁₀.

N	0.04961000	-0.03124900	2.00745900
C	0.08867600	1.13485500	2.70709300
C	0.11807400	1.16618500	4.10958000
C	0.10691300	-0.04458500	4.84122200
C	0.06662700	-1.24825800	4.09932800
C	0.03916400	-1.20379500	2.69698700
H	0.05623400	-2.20599500	4.60885600
H	0.00768800	-2.11392000	2.10784300
H	0.09548400	2.05001100	2.12464400
H	0.14872000	2.11897300	4.62759200
C	-2.79818800	1.38831600	-0.53205800
N	-1.41566400	1.53195300	-0.49168600
C	-1.14400900	2.89073700	-0.55552700
C	-2.39836800	3.63495600	-0.62832100
C	-3.41518400	2.71140200	-0.62113900
C	0.14607600	3.46799300	-0.58304300
C	1.37705000	2.77459500	-0.62742700
C	2.68880800	3.40053700	-0.76942800
C	3.61520000	2.38741800	-0.81616800
C	2.88482300	1.12638200	-0.69172000
N	1.52576000	1.39626200	-0.57581000
C	3.47932600	-0.16018700	-0.73145100
C	2.76906800	-1.38703100	-0.70478900
C	3.38516000	-2.71286500	-0.74818900
C	2.36860100	-3.63563700	-0.70220100
C	1.11587900	-2.88897000	-0.62667500
N	1.38839000	-1.52866600	-0.61919100
C	-0.17359100	-3.46589700	-0.57556000
C	-1.40520700	-2.77241200	-0.55596400
C	-2.72342900	-3.40084000	-0.55021200
C	-3.65195300	-2.38856900	-0.54017900
C	-2.91515700	-1.12496200	-0.54257400
N	-1.55085300	-1.39280300	-0.54059500
C	-3.50997200	0.16204200	-0.53330600
H	-4.47552200	2.90294100	-0.69239400
C	0.21447600	4.96748200	-0.57420300
H	2.87849400	4.46172600	-0.84937300
H	4.68319300	2.48093700	-0.94559900
C	4.97697100	-0.22599000	-0.83370700
H	4.44575300	-2.90833000	-0.80044800
H	2.46402300	-4.71222100	-0.72349600
C	-0.24160100	-4.96483400	-0.53248200

H	-2.91769600	-4.46419100	-0.55657200
H	-4.72717400	-2.48583300	-0.52492800
C	-5.01058900	0.23510200	-0.54865400
H	-2.49377200	4.70919600	-0.70199700
Zn	-0.00325600	-0.00243200	-0.20438200
C	0.23611000	5.68488100	0.63378300
C	0.30051300	7.08253700	0.68126000
C	0.34559900	7.80884800	-0.51468300
C	0.32564000	7.13053000	-1.73895500
C	0.26087400	5.73067500	-1.75269600
F	0.19432600	4.99369100	1.83656800
F	0.32056000	7.74681900	1.89234800
F	0.40982300	9.18629700	-0.48588800
F	0.36959900	7.84383200	-2.92057300
F	0.24119800	5.09238500	-2.98033500
C	-0.18846500	-5.66026700	0.68725000
C	-0.25102900	-7.05660900	0.76454900
C	-0.37157800	-7.80444400	-0.41280200
C	-0.42768200	-7.14853600	-1.64812500
C	-0.36250700	-5.74931200	-1.69156700
F	-0.19605200	-7.69869100	1.98643200
F	-0.07191800	-4.94686100	1.87220900
F	-0.43524800	-9.18097500	-0.35473900
F	-0.54663100	-7.88325200	-2.81133500
F	-0.41962900	-5.13384100	-2.92961500
C	-5.74620000	-0.21280700	-1.67111400
C	-7.14733900	-0.13849800	-1.69080100
C	-7.84253900	0.38062000	-0.57699700
C	-7.11821700	0.82755700	0.54969500
C	-5.71781300	0.75893600	0.55942300
C	5.79199100	0.21960400	0.23328200
C	7.19117700	0.15720300	0.14669800
C	7.80405300	-0.34680000	-1.02095000
C	7.00000400	-0.78795900	-2.09470400
C	5.60229000	-0.73110400	-1.99839000
H	-5.16503100	1.09697400	1.43176600
H	-7.66537700	1.22095400	1.40155800
C	-9.32699600	0.47634700	-0.54633400
H	-7.70689100	-0.47416800	-2.55727100
H	-5.21377000	-0.60246400	-2.53450600
H	4.98836300	-1.06162000	-2.83177200
H	7.48381700	-1.16681100	-2.99035800
C	9.28256300	-0.43030400	-1.16421000
H	7.81245900	0.49141100	0.97062400
H	5.32302400	0.60049800	1.13658300
O	-9.99641400	0.92483300	0.40474300
O	-9.91490100	0.00639000	-1.70771400
C	-11.38538800	0.06160700	-1.77407900

H	-11.63797400	-0.34319000	-2.75509200
H	-11.82218800	-0.54332400	-0.97322100
O	9.88181100	-0.86588500	-2.16639300
O	9.95356500	0.03494500	-0.04601300
C	11.42518100	-0.00756100	-0.09410900
H	11.74981000	0.39106100	0.86810600
H	11.76900500	-1.03766400	-0.23095900
H	-11.72934000	1.09582500	-1.67346200
H	11.79360100	0.60807400	-0.92075100
B	0.13638200	-0.05180800	6.40412100
O	0.17469100	1.17236400	7.04479200
O	0.12170700	-1.28155700	7.03458500
H	0.13550000	-1.40515200	7.99867300
H	0.19338400	1.28654800	8.00998300

Table S3-13. Atomic coordinates of L1-anatase101-M.

Ti	0.63392	15.30531	8.45636
Ti	3.44428	0.08972	7.84637
Ti	5.86697	1.97745	8.47463
Ti	8.68476	2.03056	7.89409
Ti	0.61201	3.91279	8.46258
Ti	0.66466	7.73907	8.50505
Ti	0.65200	11.50339	8.50068
Ti	3.41390	3.85056	7.84803
Ti	3.47901	7.74540	7.86787
Ti	3.47052	11.49696	7.89325
Ti	5.76548	5.73618	8.39093
Ti	5.88083	9.60983	8.73694
Ti	5.89946	13.40451	8.48108
Ti	8.71888	5.87602	7.85821
Ti	8.65286	9.64757	7.85534
Ti	8.70799	13.41392	7.89856
O	10.56332	2.04079	8.69353
O	3.85735	0.02282	5.74898
O	2.27038	15.34220	9.30666
O	1.59137	15.24387	6.90091
O	5.32297	15.30657	8.70379
O	9.10395	1.94856	5.77415
O	7.49906	2.03931	9.34539
O	3.88943	1.97926	7.82948
O	6.84180	1.95362	6.92128
O	9.12118	15.32555	7.92779
O	10.55335	5.81728	8.70492
O	10.62106	9.62047	8.63336
O	10.58386	13.43305	8.70684
O	3.85291	3.84013	5.74601
O	3.85726	7.63694	5.82569
O	3.85871	11.42616	5.82936

O	2.25176	3.90475	9.31444
O	2.28527	7.71979	9.36018
O	2.27971	11.53018	9.36918
O	1.57656	3.84226	6.90681
O	1.58951	7.64159	6.94392
O	1.60967	11.51328	6.95017
O	5.29851	3.85915	8.71723
O	5.32874	7.64347	8.51080
O	5.37930	11.52467	8.61683
O	9.11158	5.74546	5.82866
O	9.13154	9.53122	5.83184
O	9.11208	13.33242	5.80921
O	7.35865	6.07978	9.44532
O	7.63130	9.64139	9.39130
O	7.51526	13.43680	9.36801
O	3.86195	5.74479	7.84470
O	3.84832	9.60928	7.94212
O	3.88315	13.40652	7.93072
O	6.86495	5.63703	6.95073
O	6.86345	9.54734	7.02058
O	6.86285	13.42038	6.93782
O	9.07369	3.94340	7.95138
O	9.11805	7.70239	7.94777
O	9.13344	11.53154	7.93574
O	6.43325	7.52053	11.40843
C	5.68446	8.55532	11.59494
C	5.17367	8.95269	12.95850
O	5.29919	9.36187	10.60035
C	5.27922	8.07480	14.06957
C	4.58099	10.22874	13.16619
C	4.80924	8.51213	15.33083
H	5.72446	7.07090	13.94353
C	4.16169	10.58228	14.47204
H	4.46292	10.93409	12.32265
H	4.88038	7.85193	16.21712
H	3.73124	11.58494	14.67241
N	4.26493	9.75138	15.55322
H	7.14026	6.76225	10.26797
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841
Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667

Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583
O	6.76373	-0.03578	4.87464
O	5.36212	1.89800	4.08511
O	8.34378	1.89800	3.28133
O	1.52660	5.69312	4.91841
O	1.53166	9.49000	4.98906
O	1.52660	13.28688	4.91841
O	0.09219	3.83007	4.08191
O	0.12587	7.62602	4.12486
O	0.12587	11.35398	4.12486
O	3.11924	3.82597	3.25583
O	3.12760	7.62092	3.32997
O	3.12760	11.35908	3.32997
O	6.76373	3.83178	4.87464
O	6.76248	7.62102	4.98540
O	6.76248	11.35898	4.98540
O	5.36629	5.74107	4.08249
O	5.39119	9.49000	4.07365
O	5.36629	13.23893	4.08249
O	8.37288	5.74079	3.29848
O	8.37941	9.49000	3.33900
O	8.37288	13.23921	3.29848

Table S3-14. Atomic coordinates of L1-anatase101-B.

Ti	0.59251	15.24320	8.41083
Ti	3.40345	0.09139	7.84939
Ti	5.85937	1.94019	8.43619
Ti	8.67647	1.96357	7.84203
Ti	0.62177	3.86437	8.40493
Ti	0.64453	7.56442	8.48308
Ti	0.63122	11.47311	8.49252
Ti	3.45593	3.84465	7.82444
Ti	3.44429	7.54031	7.86523
Ti	3.41433	11.53492	7.87463
Ti	5.82141	5.71955	8.90880
Ti	5.61227	9.53660	8.46674
Ti	5.85485	13.37057	8.46207
Ti	8.56658	5.71537	7.81617
Ti	8.85532	9.51408	7.78540
Ti	8.67337	13.33935	7.90340

O	10.55787	1.94297	8.60677
O	3.86187	0.03069	5.74608
O	2.21541	15.28208	9.29523
O	1.57667	15.21772	6.87360
O	5.28741	15.24013	8.67764
O	9.10170	1.92494	5.76365
O	7.48567	1.94428	9.29936
O	3.83344	1.94042	7.83067
O	6.82857	1.91618	6.88190
O	9.08430	15.22465	7.85705
O	10.57850	5.72457	8.62666
O	10.69800	9.54076	8.52753
O	10.54590	13.35582	8.72322
O	3.86062	3.82449	5.74494
O	3.85930	7.64069	5.82779
O	3.85581	11.40597	5.83278
O	2.25271	3.84145	9.27686
O	2.26292	7.62442	9.35639
O	2.25711	11.53003	9.36752
O	1.59413	3.83761	6.86220
O	1.60019	7.54578	6.93263
O	1.57731	11.51547	6.93883
O	5.32657	3.78059	8.60836
O	5.34336	7.64631	8.52681
O	5.29776	11.49816	8.60581
O	9.13114	5.72870	5.77074
O	9.12159	9.53471	5.84596
O	9.10760	13.28325	5.80972
O	7.62813	5.74870	9.40000
O	7.47077	13.33970	9.36509
O	3.85166	5.72036	7.90384
O	3.79501	9.55374	7.93698
O	3.82847	13.37344	7.85465
O	6.85820	5.69493	6.97417
O	6.93832	9.53452	7.17484
O	6.83772	13.34926	6.93317
O	9.09188	3.83176	7.86149
O	9.11161	7.65120	7.94487
O	9.13075	11.44321	7.94204
O	8.39211	9.60328	9.86086
C	7.43651	9.62631	10.74038
C	7.69128	9.65764	12.20583
O	6.16541	9.61571	10.36233
C	9.00844	9.66812	12.73368
C	6.58980	9.65811	13.09562
C	9.15615	9.67310	14.13841
H	9.88393	9.66884	12.06397
C	6.84583	9.66430	14.48340

H	5.55592	9.65073	12.70458
H	10.16739	9.67735	14.59351
H	6.01234	9.66280	15.21094
N	8.10902	9.66976	15.02218
O	5.21494	5.73951	10.61450
H	5.35141	5.55163	11.57487
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841
Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667
Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583
O	6.76373	-0.03578	4.87464
O	5.36212	1.89800	4.08511
O	8.34378	1.89800	3.28133
O	1.52660	5.69312	4.91841
O	1.53166	9.49000	4.98906
O	1.52660	13.28688	4.91841
O	0.09219	3.83007	4.08191
O	0.12587	7.62602	4.12486
O	0.12587	11.35398	4.12486
O	3.11924	3.82597	3.25583
O	3.12760	7.62092	3.32997
O	3.12760	11.35908	3.32997
O	6.76373	3.83178	4.87464
O	6.76248	7.62102	4.98540
O	6.76248	11.35898	4.98540
O	5.36629	5.74107	4.08249
O	5.39119	9.49000	4.07365
O	5.36629	13.23893	4.08249
O	8.37288	5.74079	3.29848
O	8.37941	9.49000	3.33900
O	8.37288	13.23921	3.29848

Table S3-15. Atomic coordinates of L2-anatase101-M.

Ti	0.67260	15.29982	8.44654
Ti	3.48307	0.12564	7.83505
Ti	5.89437	2.01287	8.46763
Ti	8.72350	2.02513	7.87364
Ti	0.65490	3.91793	8.44483
Ti	0.66646	7.73729	8.50298
Ti	0.68296	11.50035	8.51130
Ti	3.46286	3.91083	7.82476
Ti	3.54235	7.75659	7.82201
Ti	3.50221	11.51254	7.90221
Ti	5.86029	5.76854	8.42360
Ti	5.92162	9.64563	8.78992
Ti	5.92457	13.43012	8.48521
Ti	8.71954	5.85367	7.92260
Ti	8.68314	9.63598	7.85718
Ti	8.73869	13.41028	7.89675
O	10.61379	2.03567	8.64616
O	3.87275	0.03121	5.74808
O	2.30925	15.34851	9.29293
O	1.62816	15.22873	6.89299
O	5.35193	15.33313	8.70057
O	9.11077	1.94690	5.77209
O	7.53299	2.03272	9.32219
O	3.92042	2.02482	7.81852
O	6.86257	1.95479	6.91399
O	9.15199	15.31917	7.92283
O	10.59935	5.82415	8.71176
O	10.64898	9.62120	8.70444
O	10.62891	13.42959	8.70257
O	3.86505	3.85462	5.74308
O	3.86495	7.65038	5.82611
O	3.87287	11.43388	5.82826
O	2.29555	3.94733	9.29065
O	2.34067	7.72274	9.37677
O	2.31521	11.52218	9.37279
O	1.60147	3.84233	6.88553
O	1.61537	7.64087	6.95826
O	1.63951	11.48428	6.95604
O	5.33138	3.90360	8.70237
O	5.33389	7.65854	8.65646
O	5.40893	11.53859	8.62991
O	9.12603	5.73922	5.81861
O	9.13331	9.53695	5.83161
O	9.12264	13.33565	5.80964
O	7.45495	5.95630	9.39511
O	7.65041	9.64778	9.40834
O	7.54406	13.43847	9.36769
O	3.89974	5.79212	7.83333

O	3.90366	9.61639	7.93622
O	3.91135	13.42428	7.93291
O	6.89816	5.72529	6.93818
O	6.88212	9.51933	7.04922
O	6.89033	13.42062	6.94124
O	9.12158	3.93789	7.94255
O	9.17623	7.72235	7.94547
O	9.16776	11.52588	7.93461
O	3.67071	7.30572	11.41242
C	4.96741	8.49140	13.52900
O	5.26384	9.52927	10.71461
C	4.07985	7.63862	14.24163
C	5.67570	9.49789	14.22423
C	3.95071	7.82595	15.63400
H	3.50178	6.86391	13.71363
C	5.46104	9.60728	15.61479
H	6.36910	10.18902	13.70357
H	3.27883	7.17228	16.22130
H	5.99072	10.38904	16.20784
N	4.62093	8.79823	16.33151
O	6.49322	7.13262	11.63038
H	6.90396	6.76382	10.74291
P	5.10977	8.17561	11.70676
H	3.05634	7.53161	10.53463
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841
Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667
Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583
O	6.76373	-0.03578	4.87464
O	5.36212	1.89800	4.08511
O	8.34378	1.89800	3.28133
O	1.52660	5.69312	4.91841
O	1.53166	9.49000	4.98906

O	1.52660	13.28688	4.91841
O	0.09219	3.83007	4.08191
O	0.12587	7.62602	4.12486
O	0.12587	11.35398	4.12486
O	3.11924	3.82597	3.25583
O	3.12760	7.62092	3.32997
O	3.12760	11.35908	3.32997
O	6.76373	3.83178	4.87464
O	6.76248	7.62102	4.98540
O	6.76248	11.35898	4.98540
O	5.36629	5.74107	4.08249
O	5.39119	9.49000	4.07365
O	5.36629	13.23893	4.08249
O	8.37288	5.74079	3.29848
O	8.37941	9.49000	3.33900
O	8.37288	13.23921	3.29848

Table S3-16. Atomic coordinates of L2-anatase101-B.

Ti	0.62956	15.27097	8.48256
Ti	3.41722	0.16410	7.84636
Ti	5.90729	2.02672	8.41354
Ti	8.73194	1.99076	7.81988
Ti	0.64764	3.85780	8.40630
Ti	0.57320	7.61201	8.46331
Ti	0.51466	11.45891	8.40336
Ti	3.46022	3.89758	7.82585
Ti	3.38915	7.58600	7.88941
Ti	3.46633	11.58300	7.72717
Ti	5.88857	5.78077	8.92058
Ti	5.61287	9.60529	8.44631
Ti	5.87848	13.45717	8.40230
Ti	8.58355	5.67826	7.77352
Ti	8.67241	9.52342	7.78928
Ti	8.68122	13.40156	7.83606
O	10.56005	1.94190	8.67873
O	3.86021	-0.05734	5.74942
O	2.24895	15.33645	9.35132
O	1.58792	15.23961	6.91857
O	5.33887	15.30963	8.66877
O	9.10232	1.86705	5.74595
O	7.52176	2.03010	9.27238
O	3.88046	1.95986	7.80755
O	6.85790	1.91999	6.86877
O	9.14766	15.21537	7.82101
O	10.60112	5.73564	8.60113
O	10.51467	9.60997	8.72595
O	10.54953	13.32890	8.72868
O	3.86678	3.83327	5.73079

O	3.79709	7.63915	5.84317
O	3.82598	11.45547	5.82490
O	2.26421	3.80142	9.28140
O	2.16872	7.72943	9.32464
O	2.24275	11.51266	9.40099
O	1.60391	3.81009	6.84743
O	1.48881	7.63883	6.89955
O	1.55985	11.50105	6.96358
O	5.31904	3.83620	8.52836
O	5.29815	7.70624	8.39842
O	5.31600	11.54703	8.35339
O	9.12960	5.75712	5.77388
O	9.12743	9.50753	5.84086
O	9.10728	13.31624	5.79950
O	7.64644	5.28445	9.27585
O	7.48185	13.32021	9.29724
O	3.70190	5.76180	7.85030
O	3.72244	9.64237	7.88620
O	3.81688	13.44135	7.93389
O	6.87398	5.76866	6.96247
O	6.87482	9.52212	6.99309
O	6.84420	13.43897	6.86800
O	9.21510	3.80208	7.69612
O	9.00446	7.65449	8.03794
O	9.15568	11.48454	7.78212
O	5.41562	9.55171	10.46207
C	8.08082	9.30498	12.28672
O	7.56431	9.84850	9.35905
C	7.68361	9.22829	13.61260
C	9.45941	9.24474	11.94734
C	8.64789	9.02458	14.60517
H	6.60375	9.26221	13.91781
C	10.38184	9.00124	13.03071
H	9.81418	9.31226	10.93145
H	8.37168	8.92854	15.64471
H	11.40736	8.87088	12.77983
N	9.99791	8.85853	14.30846
O	7.19853	7.31440	10.58500
O	4.98421	5.85875	10.47521
H	4.91967	6.37245	11.32932
P	7.00383	8.91734	10.77758
H	2.63456	11.29366	10.28184
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841

Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667
Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583
O	6.76373	-0.03578	4.87464
O	5.36212	1.89800	4.08511
O	8.34378	1.89800	3.28133
O	1.52660	5.69312	4.91841
O	1.53166	9.49000	4.98906
O	1.52660	13.28688	4.91841
O	0.09219	3.83007	4.08191
O	0.12587	7.62602	4.12486
O	0.12587	11.35398	4.12486
O	3.11924	3.82597	3.25583
O	3.12760	7.62092	3.32997
O	3.12760	11.35908	3.32997
O	6.76373	3.83178	4.87464
O	6.76248	7.62102	4.98540
O	6.76248	11.35898	4.98540
O	5.36629	5.74107	4.08249
O	5.39119	9.49000	4.07365
O	5.36629	13.23893	4.08249
O	8.37288	5.74079	3.29848
O	8.37941	9.49000	3.33900
O	8.37288	13.23921	3.29848

Table S3-17. Atomic coordinates of L5-anatase101-M.

Ti	0.69634	15.29435	8.46942
Ti	3.50374	0.10208	7.84394
Ti	5.93111	1.98679	8.45990
Ti	8.76003	2.03517	7.88084
Ti	0.70147	3.91159	8.47633
Ti	0.72852	7.73237	8.51355
Ti	0.65668	11.48009	8.49204
Ti	3.49962	3.88970	7.84343
Ti	3.59613	7.69574	7.87630
Ti	3.65617	11.45266	7.82027
Ti	5.96410	5.75102	8.49110
Ti	5.97392	9.57606	8.91117

Ti	5.96448	13.43162	8.48523
Ti	8.78609	5.83334	7.88288
Ti	8.69097	9.58800	7.82097
Ti	8.77941	13.42181	7.87915
O	10.64122	2.02319	8.70050
O	3.87451	0.01823	5.74706
O	2.34292	15.34797	9.30885
O	1.64342	15.22330	6.91167
O	5.38728	15.31168	8.70062
O	9.12455	1.94079	5.76836
O	7.57563	2.05027	9.31727
O	3.94494	1.98167	7.81626
O	6.90355	1.93472	6.91149
O	9.18747	15.30879	7.91439
O	10.68190	5.81582	8.68938
O	10.68499	9.61889	8.71567
O	10.66401	13.41385	8.72243
O	3.87445	3.84918	5.74319
O	3.88327	7.63702	5.82638
O	3.87835	11.42572	5.83484
O	2.34654	3.92657	9.31344
O	2.39944	7.84329	9.39918
O	2.38042	11.33502	9.49124
O	1.64564	3.84562	6.91231
O	1.68030	7.64103	6.97231
O	1.69870	11.52338	7.01274
O	5.38634	3.85445	8.69842
O	5.42736	7.63068	8.64150
O	5.45180	11.53892	8.60770
O	9.13102	5.73639	5.80903
O	9.14876	9.53261	5.82264
O	9.12714	13.33667	5.80466
O	7.59993	5.84181	9.35335
O	7.74729	9.64902	9.40776
O	7.60195	13.41840	9.34771
O	3.93666	5.76413	7.91581
O	3.99295	9.58965	7.92727
O	3.93560	13.38940	7.94017
O	6.91757	5.72485	6.94028
O	6.93190	9.53997	7.04235
O	6.92006	13.43371	6.93611
O	9.19074	3.92083	7.92172
O	9.24818	7.70834	7.92647
O	9.26233	11.51744	7.86155
O	3.05038	9.41670	11.47258
C	4.98242	9.82957	13.17502
O	5.26502	9.62524	10.60306
C	4.06557	9.85602	14.26907

C	6.36262	9.98061	13.50303
C	4.54071	10.04061	15.58356
H	2.99041	9.72679	14.06751
C	6.75004	10.14492	14.85146
H	7.14296	9.96229	12.71236
H	3.82599	10.07236	16.43570
H	7.82191	10.25307	15.13622
N	5.86719	10.18270	15.90624
B	4.44406	9.60463	11.72092
H	2.81053	8.76160	10.70396
H	2.63076	10.79741	10.33781
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841
Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667
Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583
O	6.76373	-0.03578	4.87464
O	5.36212	1.89800	4.08511
O	8.34378	1.89800	3.28133
O	1.52660	5.69312	4.91841
O	1.53166	9.49000	4.98906
O	1.52660	13.28688	4.91841
O	0.09219	3.83007	4.08191
O	0.12587	7.62602	4.12486
O	0.12587	11.35398	4.12486
O	3.11924	3.82597	3.25583
O	3.12760	7.62092	3.32997
O	3.12760	11.35908	3.32997
O	6.76373	3.83178	4.87464
O	6.76248	7.62102	4.98540
O	6.76248	11.35898	4.98540
O	5.36629	5.74107	4.08249
O	5.39119	9.49000	4.07365
O	5.36629	13.23893	4.08249

O	8.37288	5.74079	3.29848
O	8.37941	9.49000	3.33900
O	8.37288	13.23921	3.29848

Table S3-18. Atomic coordinates of L5-anatase101-M.

Ti	0.60572	15.24983	8.44490
Ti	3.40829	0.07530	7.84866
Ti	5.86034	1.93649	8.43678
Ti	8.68787	1.98386	7.85770
Ti	0.62267	3.88824	8.41559
Ti	0.61957	7.60058	8.49493
Ti	0.51022	11.40522	8.41002
Ti	3.44789	3.84728	7.81439
Ti	3.44370	7.57195	7.87235
Ti	3.53110	11.48011	7.77734
Ti	5.76510	5.69568	8.95729
Ti	5.68693	9.49323	8.67401
Ti	5.87243	13.38155	8.47671
Ti	8.53369	5.73562	7.81093
Ti	8.64488	9.48580	7.82459
Ti	8.68194	13.36775	7.88115
O	10.57113	1.95063	8.60870
O	3.85495	-0.03670	5.74889
O	2.23483	15.23991	9.31570
O	1.57718	15.22824	6.89483
O	5.29255	15.23615	8.69714
O	9.10642	1.92529	5.77083
O	7.48714	1.94726	9.30134
O	3.84120	1.92213	7.82273
O	6.84037	1.90803	6.88525
O	9.09429	15.23027	7.91994
O	10.58715	5.74364	8.63349
O	10.57560	9.54856	8.71102
O	10.55154	13.32958	8.70906
O	3.85583	3.83361	5.74342
O	3.84321	7.63801	5.82744
O	3.85889	11.43142	5.83714
O	2.25723	3.84105	9.27883
O	2.25739	7.63985	9.36251
O	2.24198	11.44636	9.39899
O	1.58452	3.83562	6.86384
O	1.58176	7.63200	6.94470
O	1.60264	11.43278	6.96002
O	5.30774	3.76567	8.60324
O	5.32615	7.63753	8.58552
O	5.29980	11.51091	8.60901
O	9.13728	5.73188	5.78855

O	9.12153	9.52319	5.84302
O	9.10203	13.32991	5.80592
O	7.54950	5.74573	9.39632
O	7.49285	13.32974	9.35683
O	3.81389	5.72482	7.89556
O	3.86181	9.56138	7.94819
O	3.81392	13.34873	7.92155
O	6.84677	5.68956	6.95960
O	6.86931	9.53977	7.07448
O	6.83428	13.35057	6.92950
O	9.09328	3.84483	7.90640
O	9.09090	7.65176	7.94266
O	9.14741	11.44313	7.83704
O	7.70396	9.63538	9.50934
C	7.79503	9.77822	12.15101
O	5.67125	9.64027	10.60447
C	9.21425	9.80879	12.23801
C	7.06924	9.83227	13.37440
C	9.83245	9.89831	13.50390
H	9.83145	9.75323	11.31876
C	7.77258	9.91996	14.59911
H	5.96354	9.80308	13.36877
H	10.93967	9.92134	13.59199
H	7.22435	9.96180	15.56052
N	9.14170	9.95765	14.68611
O	4.99314	5.73807	10.57826
H	4.28858	5.84652	11.26782
H	2.54247	11.53158	10.34083
B	7.05109	9.68663	10.79381
Ti	1.95273	-0.03160	4.88410
Ti	4.87400	-0.02149	4.19308
Ti	7.19495	1.89800	4.92578
Ti	10.10811	1.89800	4.19733
Ti	1.95273	3.82760	4.88410
Ti	1.96603	7.62556	4.95841
Ti	1.96603	11.35444	4.95841
Ti	4.87400	3.81749	4.19308
Ti	4.88673	7.59140	4.26667
Ti	4.88673	11.38860	4.26667
Ti	7.20627	5.69554	4.92413
Ti	7.20629	9.49000	4.96097
Ti	7.20627	13.28446	4.92413
Ti	10.12802	5.69422	4.23725
Ti	10.13103	9.49000	4.28219
Ti	10.12802	13.28578	4.23725
O	1.50815	1.89800	4.85674
O	0.09219	-0.03407	4.08191
O	3.11924	-0.02997	3.25583

O 6.76373 -0.03578 4.87464
O 5.36212 1.89800 4.08511
O 8.34378 1.89800 3.28133
O 1.52660 5.69312 4.91841
O 1.53166 9.49000 4.98906
O 1.52660 13.28688 4.91841
O 0.09219 3.83007 4.08191
O 0.12587 7.62602 4.12486
O 0.12587 11.35398 4.12486
O 3.11924 3.82597 3.25583
O 3.12760 7.62092 3.32997
O 3.12760 11.35908 3.32997
O 6.76373 3.83178 4.87464
O 6.76248 7.62102 4.98540
O 6.76248 11.35898 4.98540
O 5.36629 5.74107 4.08249
O 5.39119 9.49000 4.07365
O 5.36629 13.23893 4.08249
O 8.37288 5.74079 3.29848
O 8.37941 9.49000 3.33900
O 8.37288 13.23921 3.29848

Sample input file for the geometry optimizations using SIESTA 3.1 code.

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SystemName      L1_M
NumberOfAtoms   110
NumberOfSpecies 5

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PAO.BasisSize DZ
PAO.EnergyShift 0.01 eV
PAO.SplitNorm 0.15

MeshCutoff 200.0 Ry

---Exchange-correlation functionals
XC.functional    GGA
XC.authors       PBE

ElectronicTemperature 300 K
MaxSCFIterations     30000
DM.MixingWeight      0.02
DM.NumberPulay        4
DM.Tolerance          1.0D-5
DM.UseSaveDM          T

SolutionMethod       diagon

LatticeConstant      1.0000 Ang

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0.00000000 0.00000000 20.00000000
%endblock LatticeVectors

# K-POINTS
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 0 1 0 0.0
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%endblock kgrid_Monkhorst_Pack

BandLinesScale          ReciprocalLatticeVectors

MD.TypeOfRun           cg
MD.NumCGsteps          50
MD.MaxCGDispl          0.1  Ang
MD.MaxForceTol         0.04 eV/Ang

WriteCoorXmol .true.

%block GeometryConstraints
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%endblock GeometryConstraints

%block ChemicalSpeciesLabel
 1          22 Ti
 2          8 O
 3          6 C
 4          1 H
 5          7 N
%endblock ChemicalSpeciesLabel

AtomicCoordinatesFormat Ang
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 5.86697 1.97745 8.47463 1
 8.68476 2.03056 7.89409 1
 0.61201 3.91279 8.46258 1
 0.66466 7.73907 8.50505 1
 0.65200 11.50339 8.50068 1
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 3.47901 7.74540 7.86787 1
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Complete References with more than 10 authors:

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